

BRL R 1306

# BRL

AD

REPORT NO. 1306

## CALCULATIONS FOR AIR FLOWS IN DISSOCIATION EQUILIBRIUM

by

Nathan Gerber  
Joan M. Bartos

November 1965

Distribution of this document is unlimited.

COUNTED 1A

U. S. ARMY MATERIEL COMMAND  
BALLISTIC RESEARCH LABORATORIES  
ABERDEEN PROVING GROUND, MARYLAND

BRL  
1306  
c.2

Destroy this report when it is no longer needed.  
Do not return it to the originator.

The findings in this report are not to be construed as  
an official Department of the Army position, unless  
so designated by other authorized documents.

BALLISTIC RESEARCH LABORATORIES

REPORT NO. 1306

NOVEMBER 1965

Distribution of this document is unlimited.

CALCULATIONS FOR AIR FLOWS IN  
DISSOCIATION EQUILIBRIUM

Nathan Gerber  
Joan M. Bartos

Exterior Ballistics Laboratory

RDT & E Project No. 1A222901A201

ABERDEEN PROVING GROUND, MARYLAND

TECHNICAL LIBRARY  
BLD-4 313  
ABERDEEN PROVING GROUND, MD.  
STEAP-TL

BALLISTIC RESEARCH LABORATORIES

REPORT NO. 1306

NGerber and JMBartos  
Aberdeen Proving Ground, Md.  
November 1965

CALCULATIONS FOR AIR FLOWS IN  
DISSOCIATION EQUILIBRIUM

ABSTRACT

Results of calculations carried out for a model of air in dissociation equilibrium are presented in graphical form. The quantities computed are i) flow variables (including species concentrations) behind normal and oblique shock waves, ii) flow variables in axisymmetric conical flow fields, iii) stagnation point values of flow variables on the 'stagnation' streamline behind two-dimensional and axisymmetric detached shock waves, and iv) flow variable gradients at the shock wave on stagnation streamlines. Computations are given for free stream temperatures of  $273.16^{\circ}\text{K}$  and  $300^{\circ}\text{K}$ , free stream pressures of 1.0, .1, .01, .001, and .0001 atmospheres, and a range of initial Mach numbers and cone angles to provide flow field temperatures in the range  $3000^{\circ}\text{K}$  -  $10,000^{\circ}\text{K}$ . Brief derivations of the equations employed are given.

The present calculations are oriented toward application in experiments in hypersonic flow with ground facilities such as shock tubes and ballistic ranges. In addition, they furnish important supplementary information to theoretical studies of nonequilibrium flows.

## TABLE OF CONTENTS

	Page
ABSTRACT. . . . .	3
SYMBOLS . . . . .	7
1. INTRODUCTION. . . . .	9
2. AIR MODEL . . . . .	10
3. SHOCK WAVE CALCULATIONS . . . . .	13
4. CONICAL FLOW. . . . .	14
5. STAGNATION VALUES BEHIND NORMAL SHOCKS. . . . .	15
6. GRADIENTS AT SHOCK ON STAGNATION STREAMLINE . . . . .	15
7. COMPUTATIONAL RESULTS . . . . .	17
ACKNOWLEDGEMENTS. . . . .	17
REFERENCES. . . . .	56
APPENDIX - COEFFICIENTS OF EQUATIONS (2.6) AND (2.9). . . . .	57
DISTRIBUTION LIST . . . . .	61

# SYMBOLS

$A_{kl}$	Functions occurring in Eqs (2.6) and (2.9)
$c_i$	Concentration of $i^{\text{th}}$ species [k mol/kg of mixture]
$c_p$	Specific heat of air at constant pressure [Dyn m/kg deg K] ( <u>1 Dyn = 1 Newton = <math>10^5</math> dynes</u> )
$c_{p_i}$	= $dh_i/dt$ (see Table II) [Dyn m/k mol deg K]
$c_v$	Specific heat of air at constant volume [Dyn m/kg deg K]
$h$	Specific enthalpy [Dyn m/kg] = $\sum c_i h_i$
$h_i$	Molar specific enthalpy of $i^{\text{th}}$ species (see Table II)
$k_{f_j}, k_{b_j}$	Reaction rate constants for $j^{\text{th}}$ reaction, forward and backward, respectively (see Table I)
$K_j$	( $\equiv k_{f_j} / k_{b_j}$ ). Equilibrium constant for $j^{\text{th}}$ reaction (see Table I)
$K_w$	Shock wave curvature
$M_\infty$	Free stream Mach number
$n$	Arc length along curve normal to streamlines [m, mm]
$p$	Pressure [Dyn/m <sup>2</sup> ]
$q$	Flow velocity [m/sec]
$R$	Universal gas constant = 8312.4 [Dyn m/k mol deg K]
$R_w$	Shock wave radius of curvature ( = $1/K_w$ )
$r$	Radial polar coordinate ( = $[x^2 + y^2]^{1/2}$ )
$s$	Arc length along streamline [m, mm]
$T$	Temperature [deg K]
$U$	Velocity component in $r$ direction ( = $q \cos [\theta - \phi]$ )
$V$	Velocity component in $\phi$ direction ( = $q \sin [\theta - \phi]$ )
$W_i$	Molecular weight of $i^{\text{th}}$ species [kg/k mol]
$W_\infty$	Molecular weight of air ( = 28.8587 gm/mol )
$\theta$	Angle between shock wave and x-axis [radians, deg]
$\gamma$	Ratio of specific heats of air ( $c_p/c_v$ )

$\epsilon$	= 0 for two-dimensional flow; = 1 for axisymmetric flow
$\theta$	Angle between streamline and x-axis [radians, deg]
$\rho$	Density $[\text{kg/m}^3]$
$\sigma$	Arc length along shock wave
$\varphi$	Angular polar coordinate ( $\varphi = \arctan y/x$ )

#### Subscripts

b	Body surface
i	i <sup>th</sup> species
j	j <sup>th</sup> reaction
t	Stagnation point
w	Shock wave
$\infty$	Free stream

## 1. INTRODUCTION

Current experimental research in hypersonic flow over two-dimensional and axisymmetric bodies makes it desirable to have available calculations for air in chemical (dissociation) equilibrium. Primarily these are values of the flow variables (including concentrations of chemical species) behind shock waves and in conical flow fields. Furthermore, it is desired to have these data for ranges of free stream conditions applicable to experiments with ground facilities, such as shock tubes and ballistic ranges. To this end extensive calculations of shock wave quantities and conical flows have been carried out on the BRL high-speed computers for a model of air (see Section 2) in chemical equilibrium, and results are presented graphically here in Figs. I.1 through II.9.\* Brief derivations of the equations are given in Sections 3 and 4. Although computations of the above have already been carried out by other authors (e.g., Refs. 1 through 4 and the references contained in them), the present paper furnishes information hitherto not available, to the authors' knowledge, in a form convenient for work in hypersonic ground facilities.

The present calculations also supply important supplementary information for the determination of nonequilibrium dissociating airflow over wedges and cones.<sup>5\*\*</sup> Certain features of these nonequilibrium flows (e.g., entropy layers, oblique shock equilibrium regions) require knowledge of equilibrium values.

Studies being conducted at BRL on the subsonic region in front of a supersonic blunt body by analysis of interferometric data suggest that theoretically determined information on the stagnation streamline would be useful. Therefore, extensive calculations were performed to obtain stagnation values of the flow variables, and, in addition, gradients of the flow variables along the stagnation streamline at the shock wave. These results are presented graphically in Figs. III.1 through IV.8; derivations are given in Sections 5 and 6.

---

\*The computer program is available for cases not explicitly graphed in this report should exact values be required.

\*\* Superscript numbers denote references which may be found on page 56.



## 2. AIR MODEL

The air is considered to be a mixture of neutral species consisting of O and N atoms, and  $N_2$ ,  $O_2$ , and NO molecules. This mixture of particles is assumed to be in translational, rotational, and vibrational equilibrium at all times. Effects of electronic excitation and vibration-dissociation coupling are neglected. The chemical reactions are listed in Table I.

Let  $c_i$  denote the concentration of the  $i^{\text{th}}$  species  $M_i$  (moles of  $M_i$  per unit mass of air) and  $K_j$  the equilibrium constant of the  $j^{\text{th}}$  reaction.  $K_j$  can be determined quite accurately from quantum statistical calculations, and the results are often fitted to an equation of the form

$$K_j = A_j T^{n_j} \exp(-E_j/T) ,$$

where  $T$  is temperature, and  $A_j$ ,  $n_j$ , and  $E_j$  are constants. The values listed for equilibrium constants in Table I are based on data given in Refs. 6 and 7.

The law of mass action for equilibrium flow leads to the relations

$$c_{O_2} = \frac{\rho c_O^2}{K_1} , \quad c_{N_2} = \frac{\rho c_N^2}{K_2} , \quad c_{NO} = \frac{\rho c_N c_O}{K_5} , \quad (2.1)$$

(where  $\rho$  is the density of the air) plus the following restrictions:

$$L_1(T) \equiv \frac{K_6 K_7}{K_{10}} = 1, \quad L_2(T) \equiv \frac{K_2 K_7}{K_1 K_6} = 1, \quad L_3(T) \equiv \frac{K_5 K_7}{K_1} = 1 .$$

Calculation shows that  $L_1$ ,  $L_2$ , and  $L_3 = 1 \pm .10$  for a limited temperature range, the results being valid only within this range ( $\sim 4000^\circ$  --  $8000^\circ K$ ).

The flows studied here are produced by objects moving at constant supersonic speed in stationary air, taken to be a mixture of two ideal gases  $O_2$  and  $N_2$  having concentrations

$$c_{O_2_\infty} = \frac{.21153}{W_\infty} \frac{\text{mole}}{\text{gm air}} , \quad c_{N_2_\infty} = \frac{.78847}{W_\infty} \frac{\text{mole}}{\text{gm air}}$$

where the subscript  $\infty$  denotes free stream conditions, and  $W_\infty$  is the molecular weight of air (= 28.85870).

The conservation of chemical elements leads to the following relations:

$$c_{O_2_\infty} = c_{O_2} + (1/2) c_O + (1/2) c_{NO} \quad (2.2)$$

$$c_{N_2_\infty} = c_{N_2} + (1/2) c_N + (1/2) c_{NO} .$$

Substitution of Eq. (2.1) into Eq. (2.2) gives

$$c_{O_2_\infty} = \frac{\rho c_O^2}{K_1} + (1/2) c_O + (1/2) \frac{\rho c_N c_O}{K_5} \quad (2.3)$$

$$c_{N_2_\infty} = \frac{\rho c_N^2}{K_2} + (1/2) c_N + (1/2) \frac{\rho c_N c_O}{K_5} .$$

Considering each component as an ideal gas, the equation of state for the  $i^{th}$  species is ( $p$ ,  $\rho$ , and  $T$  being pressure, density, and temperature, respectively)

$$p_i = R \rho_i T / W_i$$

and for the mixture (where  $p = \sum_i p_i$ ,  $\rho = \sum_i \rho_i$ )

$$p = R \rho T \sum_i c_i \quad (\text{with } c_i = (\rho_i / \rho) / W_i) . \quad (2.4)$$

$R$  is the universal gas constant,  $W_i$  the molecular weight of the  $i^{th}$  species,  $c_i$  the concentration. By Eq. (2.2)

$$\sum_i c_i = (1/W_\infty) + (1/2)(c_O + c_N) . \quad (2.5)$$

The differentiated versions of Eqs. (2.4) and (2.3) give the following useful set of linear equations for  $dp$ ,  $dc_O$ ,  $dc_N$ , and  $dT$ :

$$A_{11}dp + A_{12}dc_O + A_{13}dc_N + A_{14}dT = (2/p) dp \quad (2.6a)$$

$$A_{21}dp + A_{22}dc_O + A_{23}dc_N + A_{24}dT = 0 \quad (2.6b)$$

$$A_{31}dp + A_{32}dc_O + A_{33}dc_N + A_{34}dT = 0 . \quad (2.6c)$$

Expressions for the coefficients  $A_{kl}$  are presented in the Appendix.

A fourth relation, valid along streamlines, is obtained from the energy equation

$$dh = (1/\rho) dp \quad (2.7)$$

where  $h$  is enthalpy per unit mass. The total enthalpy is the sum of the enthalpies of the components:

$$h = \sum_i c_i h_i . \quad (2.8)$$

Expressions for  $h_i$  (enthalpy per mole) and  $dh_i/dT$  ( $\equiv c_{p_i}$ ) are given in Table II for all the species. Eq. (2.7) becomes

$$A_{41}dp + A_{42}dc_O + A_{43}dc_N + A_{44}dT = (1/\rho) dp , \quad (2.9)$$

the coefficients appearing in the Appendix.

Eqs. (2.1) are differentiated to give

$$dc_{O_2} = (c_O/K_1) \left[ c_O dp + 2p dc_O - (\rho c_O/K_1) (dK_1/dT) dT \right]$$

$$dc_{N_2} = (c_N/K_2) [c_N dp + 2\rho dc_N - (\rho c_N/K_2)(dK_2/dT)dT] \quad (2.10)$$

$$dc_{N_0} = \frac{1}{K_5} \left[ c_O c_N dp + \rho (c_N dc_O + c_O dc_N) - \frac{\rho c_O c_N}{K_5} \frac{dK_5}{dT} dT \right].$$

### 3. SHOCK WAVE CALCULATIONS

The conditions immediately behind a shock (denoted by the subscript w) are given by the following relations obtained from the conservation laws\* (referring to Fig. 3.1):

$$\rho_w \tan (\beta - \theta_w) = \rho_\infty \tan \beta \quad (3.1)$$

$$q_w \cos (\beta - \theta_w) = q_\infty \cos \beta \quad (3.2)$$

$$p_w = p_\infty + \rho_\infty q_\infty^2 (1 - \rho_\infty/\rho_w) \sin^2 \beta \quad (3.3)$$

$$h_w = h_\infty + (1/2) q_\infty^2 [1 - (\rho_\infty/\rho_w)^2] \sin^2 \beta \quad (3.4)$$

where  $\beta$  is the angle of inclination of the shock wave,  $q$  the flow speed, and  $\theta$  the angle of inclination of the flow. In the free stream

$$h_\infty = (7/2)RT_\infty/W_\infty, \quad p_\infty = R\rho_\infty T_\infty/W_\infty \quad (3.5)$$

$$\rho_\infty q_\infty^2 / p_\infty = \gamma_\infty M_\infty^2,$$

where  $M_\infty$  is the Mach number and  $\gamma_\infty$  is the ratio of specific heats, having the value 1.4.

The flow variables behind the shock wave are determined by solving Eqs. (3.1), ..., (3.4), together with the equation of state Eq. (2.4), plus Eqs. (2.1) and (2.3). These form a set of ten functional equations for the ten variables  $p_w$ ,  $\theta_w$ ,  $q_w$ ,  $T_w$ ,  $\rho_w$ ,  $(c_{O_2})_w$ ,  $(c_{N_2})_w$ ,  $(c_O)_w$ ,  $(c_N)_w$ , and  $(c_{NO})_w$ , when the parameters  $M_\infty$ ,  $T_\infty$ ,  $\rho_\infty$ , (or  $p_\infty$ ) and  $\beta$  are given. The system of

---

\*The basic jump conditions for a steady oblique shock wave requiring conservation of mass, momentum, and energy can be found in many places; e.g., p. 8 of Ref. 8.

equations is solved on the BRL high speed computers by successive application of the method of "regula falsi," or "false position." Frequently  $\theta_w$  is given, and  $\beta$  is the unknown quantity; an additional iterative procedure (e.g., regula falsi) can then find the  $\beta$  corresponding to a given  $\theta_w$ .

Figs. I.1 through I.11 contain curves of flow variables behind normal and oblique shock waves. Pressure, temperature, density, and species concentrations are conveniently expressible as functions of the parameter  $M_\infty \sin \beta$  for given free stream temperature and pressure. For the flow deflection  $\theta$ , another parameter,  $M_\infty$ , is required.

#### 4. CONICAL FLOW

In axisymmetric conical flow a straight cone of half angle  $\theta_b$  gives rise to a straight attached shock wave inclined at angle  $\beta$ . It is convenient here to introduce polar coordinates  $r, \varphi$  (as in Fig. 3.1.b) and to employ  $U$  and  $V$ , the components of velocity in the  $r$  and  $\varphi$  directions, respectively. The values of  $\varphi$  at the shock and body are, respectively

$$\varphi_w = \beta, \quad \theta_b = \varphi_b \quad (4.1)$$

With the condition that the flow variables be independent of radius ( $\partial/\partial r = 0$ ), the mass and momentum conservation relations reduce to

$$dU/d\varphi = V \quad (4.2.a)$$

$$dV/d\varphi = - (V/\rho) d\rho/d\varphi - [2U + V \cot \varphi] \quad (4.2.b)$$

$$dp/d\varphi = V^2 d\rho/d\varphi + \rho V[U + V \cot \varphi] \quad (4.2.c)$$

On substituting  $dp/d\varphi$  from Eq. (4.2.c) into the right hand sides of Eqs. (2.6) and (2.9) one obtains four linear algebraic equations for  $dp/d\varphi$ ,  $dc_O/d\varphi$ ,  $dc_N/d\varphi$ , and  $dT/d\varphi$ , which are solved to give differential equations

$$dp/d\varphi = F_1, \quad dc_O/d\varphi = F_2, \quad dc_N/d\varphi = F_3, \quad dT/d\varphi = F_4 \quad (4.3)$$

where  $F_1, F_2, F_3$ , and  $F_4$  are functions of  $\varphi, U, V, \rho, p, T, c_O$ , and  $c_N$ .

Eqs. (4.2), (4.3), and (2.10) form a set of ten first order differential equations for  $U, V, p, \rho, T, c_{O_2}, c_{N_2}, c_O, c_N$ , and  $c_{NO}$ , which can be

integrated numerically by the Runge-Kutta method<sup>9</sup> on the high speed computers.

The initial conditions are taken at the shock wave; for a given atmosphere, speed, and shock inclination, all quantities are known here. The terminal condition is  $V_b = 0$ . This condition makes it convenient to use  $V$  as the independent variable instead of  $\phi$ .

Figs. II.1 through II.9 contain data for conical flow, for which many properties of interest can be accurately represented as functions of the parameter  $M_\infty \sin \theta_b$ , given the free stream conditions. Pressure, temperature, density, and species concentrations on the body surface are plotted against  $M_\infty \sin \theta_b$ , the shock wave angle is also presented in this form.

## 5. STAGNATION VALUES BEHIND NORMAL SHOCKS

Calculations are made of the flow variables when the fluid is brought to rest behind a normal shock wave, as for instance, at the intersection of the axial streamline with the surface of a symmetric blunt body (point O in Fig. 3.1.a). The independent variable in eqs. (2.6), (2.9), and (2.10) is taken to be the velocity  $q$ . These equations plus the momentum relation

$$dp/dq = -\rho q$$

form a set of differential equations which are integrated numerically from  $q = q_w$  to  $q = 0$ , the terminal values of the variables giving the stagnation conditions.

Figs. III.1 through III.9 present the thermodynamic variables and species concentrations for stagnation flow behind normal shock waves as functions of  $M_\infty$ . This information is useful in studying the subsonic region between the surface of a two-dimensional or axisymmetric blunt body and the detached shock wave ahead of it.

## 6. GRADIENTS AT SHOCK ON STAGNATION STREAMLINE

The flow variable gradients along the central streamline behind a curved shock (point P in Fig. 3.1.a) can be calculated if the curvature of the shock  $K_w$  is known. If  $\sigma$  is arc length along the shock wave, it is seen

from the relation

$$\frac{d}{d\sigma} = K_w \frac{d}{d\beta} = [\cos(\beta - \theta)] \frac{\partial}{\partial s} + [\sin(\beta - \theta)] \frac{\partial}{\partial n}$$

(where  $s$  and  $n$  are arc lengths along streamlines and their orthogonal trajectory, respectively) that

$$(\partial/\partial n)_{x\text{-axis}} = K_w (d/d\beta)_\beta = 90^\circ. \quad (6.1)$$

The momentum conservation equation\*

$$\rho q^2 \partial\theta/\partial s = - \partial p/\partial n \quad (6.2)$$

shows that  $(dp/d\beta)_\beta = 90^\circ = 0$ . Then, by Eq. (2.6) and Eq. (3.4) differentiated with respect to  $\beta$ ,  $dp/d\beta = dT/d\beta = dc_O/d\beta = dc_N/d\beta = 0$  at the  $x$ -axis. By the differentiated Eq. (3.1) and Eq. (6.1)

$$(\partial\theta/\partial n)_\beta = 90^\circ = [1 - \rho/\rho_\infty] K_w. \quad (6.3)$$

Expanding  $[(\sin \theta)/y]$  near the  $x$ -axis, noting that  $dy_w = d\sigma \sin \beta$ ,

$$(\sin \theta)/y_w = [d\theta/dy]_{y_w = 0} y_w + \dots / y_w = (d\theta/d\sigma)_{y_w = 0} + \dots$$

Therefore

$$(\sin \theta)/y_w \cong K_w (d\theta/d\beta)_\beta = 90^\circ = [1 - (\rho/\rho_\infty)] K_w. \quad (6.4)$$

Substituting Eqs. (6.3) and (6.4) into the flow equation

$$\frac{1}{\rho} \frac{\partial p}{\partial s} - \frac{1}{\rho q^2} \frac{\partial p}{\partial s} + \frac{\partial \theta}{\partial n} + \epsilon \frac{\sin \theta}{y} = 0 \quad (6.5)$$

where  $\epsilon = 0$  and  $1$  for two-dimensional and axisymmetric flow, respectively, one obtains

$$dp/ds = q^2 dp/ds - (1 + \epsilon) K_w \rho q^2 [(\rho/\rho_\infty) - 1]. \quad (6.6)$$

On substituting  $dp/ds$  from Eq. (6.6) into Eq. (2.6) the gradients of the flow variables are determined. It is seen that the gradients are all proportional to  $K_w$ , and that for a given  $M_\infty$  the axisymmetric gradients are

---

\*Eqs. (6.2) and (6.5) expressing conservation of mass and momentum are found, e.g., in Section 3 of Ref. 5.

twice those for two-dimensional flow.

Gradients along the stagnation streamline at the shock wave are shown in Figs. IV.1 through IV.8 for the thermodynamic variables and the species concentrations. Arc length is given in terms of the radius of curvature of the shock wave,  $R_w$ , at the x-axis.

## 7. COMPUTATIONAL RESULTS

Computational results are presented in the diagrams which follow. Graphs of desired quantities are plotted for free stream temperatures of 273.16°K and 300°K, and free stream pressures of 1.0, .1, .01, .001, and .0001 atmospheres; the choice of Mach numbers, cone angles and shock wave angles makes possible a temperature range coverage of 3,000°K to 10,000°K. A complete survey for air in dissociation equilibrium is not feasible because of the multitude of combinations of parameters. It is felt, nevertheless, that the following set of diagrams can be used to obtain approximate information adequate for planning experiments, predicting and checking experimental results over a wide range of conditions attainable in the laboratory.

Comparisons were made of the present results with previously published results in which more accurate models of high temperature air were assumed. For example, conical flow parameters (Fig. II.1 - - II.4) practically coincide with those of Romig (Ref. 4). A comparison of shock and stagnation pressure calculations with those of Feldman (Ref. 1) is shown in Fig. 7.1; the largest discrepancy of all the parameters is found in the stagnation pressure.

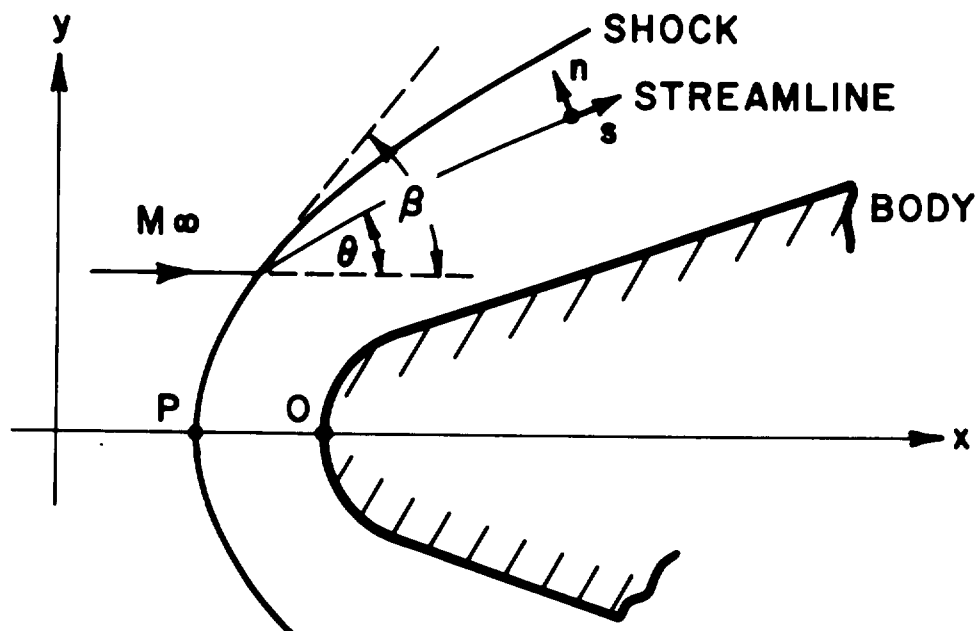
## ACKNOWLEDGEMENTS

The authors wish to express their appreciation to the following persons: Barbara Bilsborough, for programming the calculations of shock wave quantities and conical flows; Donald Taylor, for assistance on computational problems; Joseph Spurk, for assistance on questions concerning aerodynamical theory; William Hammond and Vernon Mackey, for preparing the diagrams in this report.

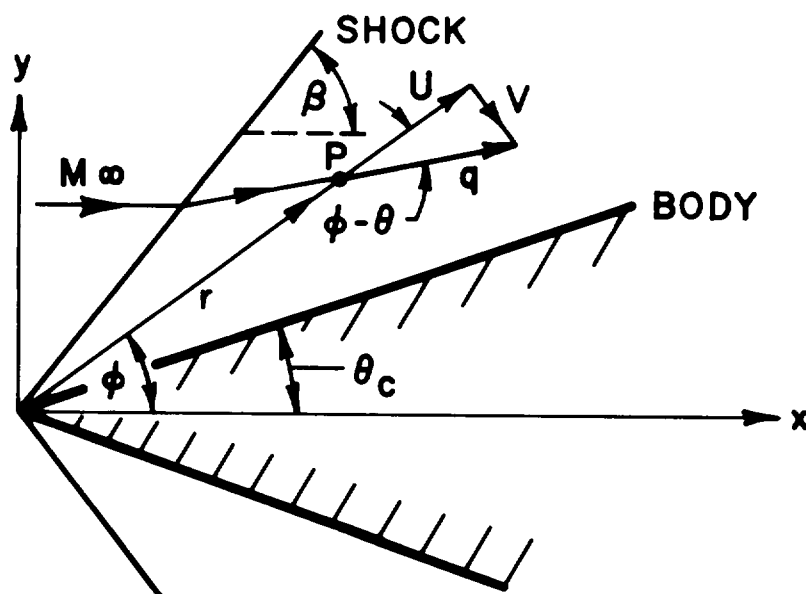
NATHAN GERBER

JOAN M. BARTOS





a. DETACHED SHOCK



b. ATTACHED SHOCK

FIG. 3.1. CROSS-SECTION DIAGRAM OF FLOW FIELD

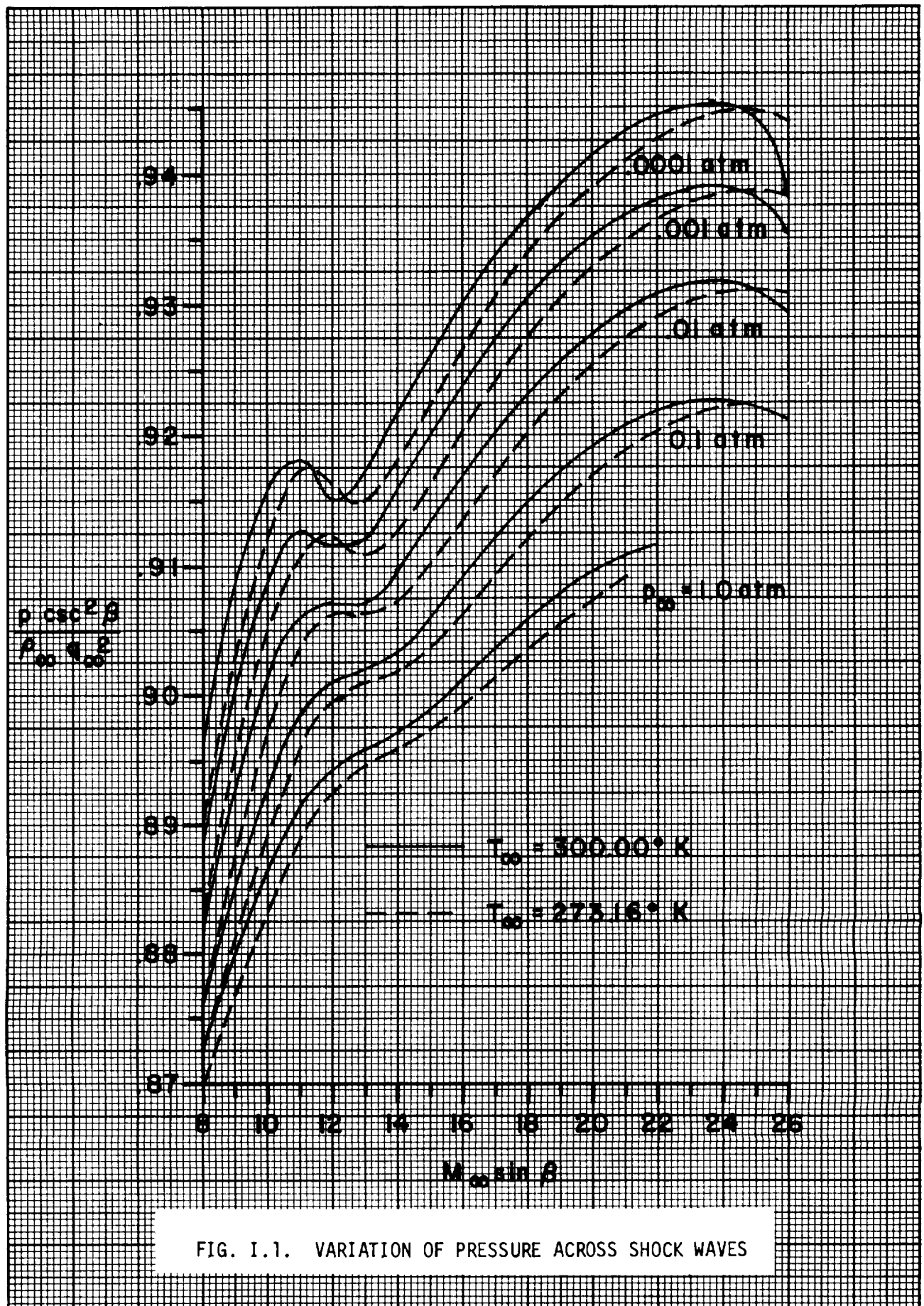


FIG. I.1. VARIATION OF PRESSURE ACROSS SHOCK WAVES

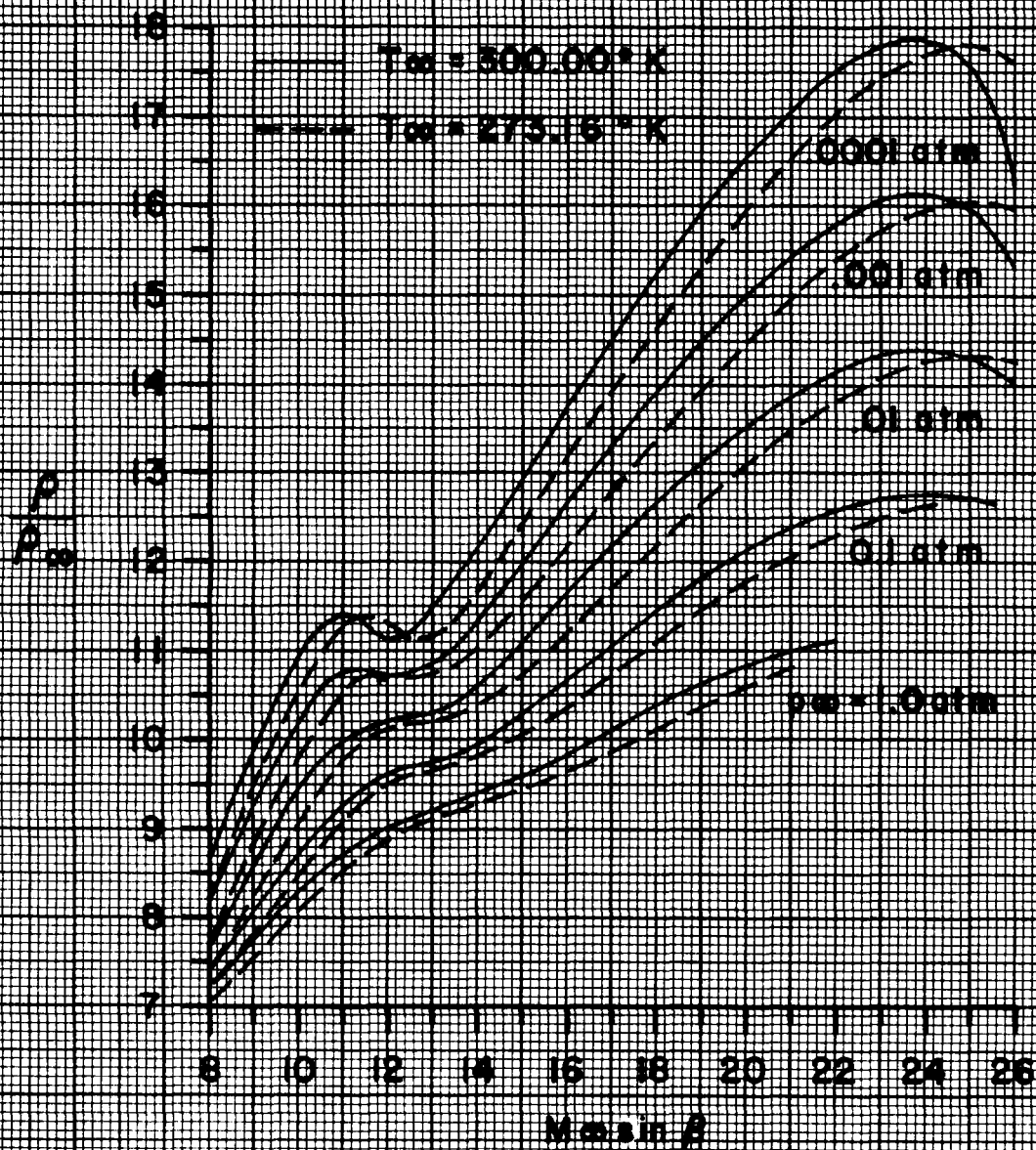


FIG. 1.2. VARIATION OF DENSITY ACROSS SHOCK WAVES

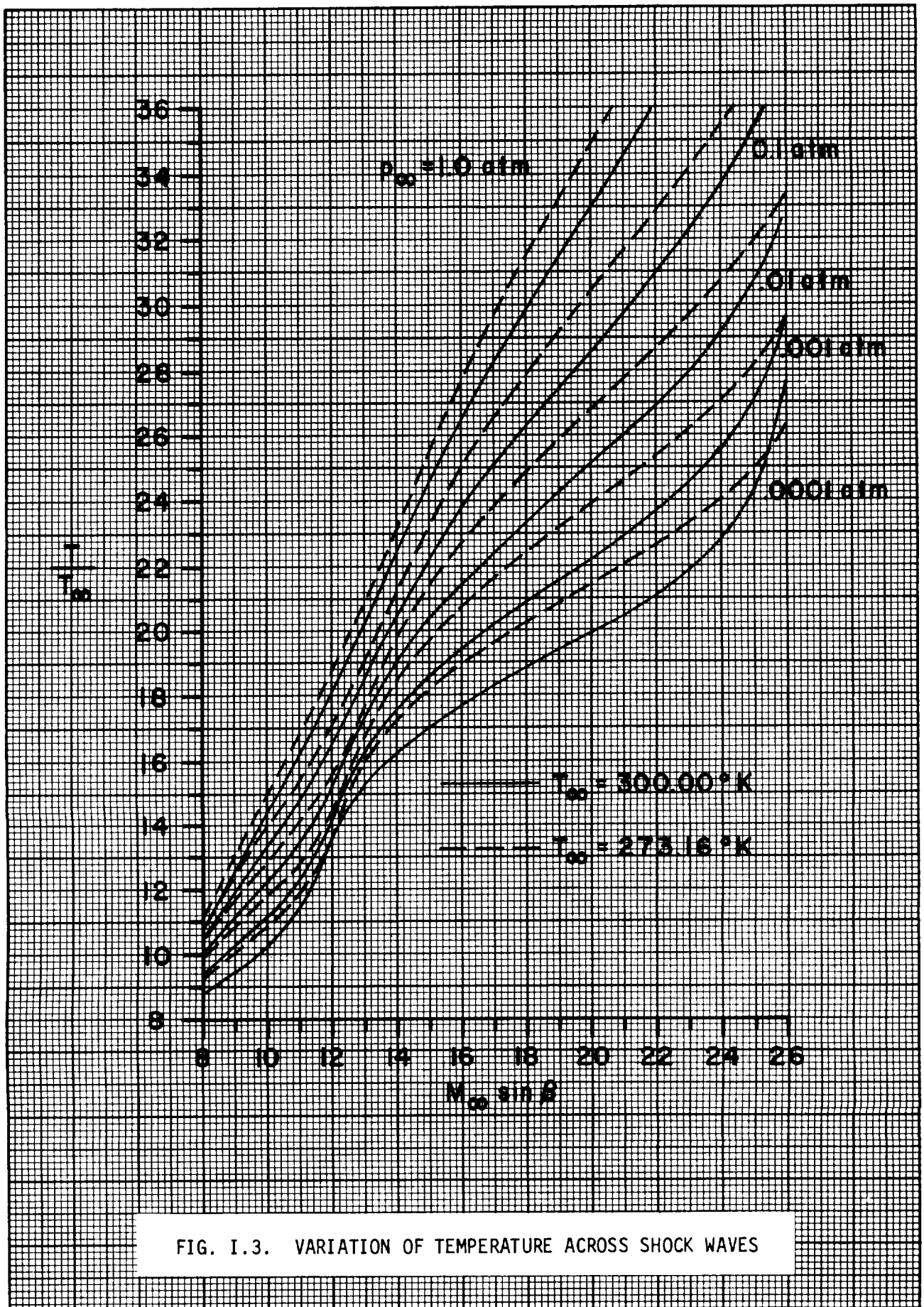


FIG. I.3. VARIATION OF TEMPERATURE ACROSS SHOCK WAVES

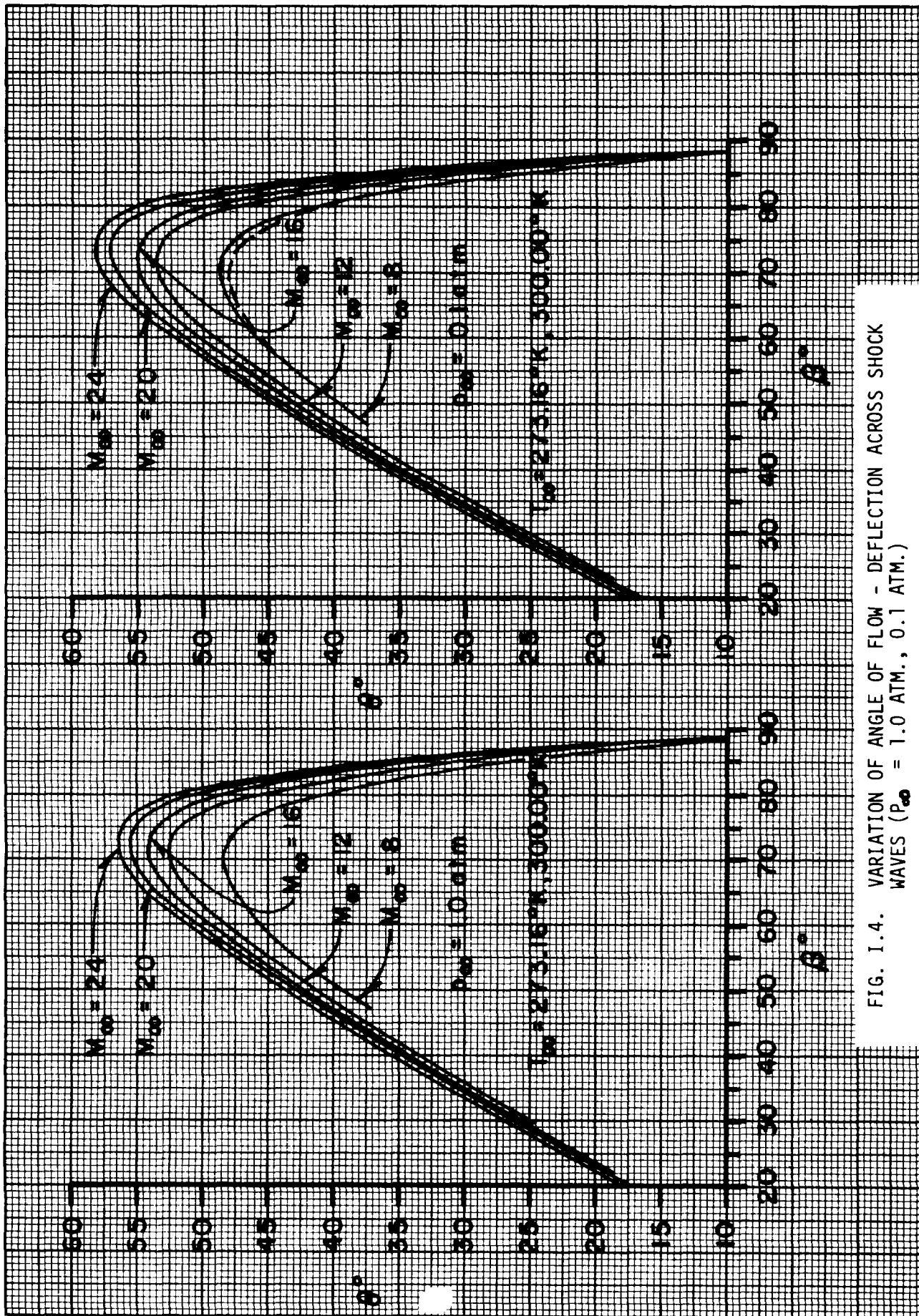


FIG. I.4. VARIATION OF ANGLE OF FLOW - DEFLECTION ACROSS SHOCK WAVES ( $P_\infty = 1.0 \text{ ATM.}, 0.1 \text{ ATM.}$ )



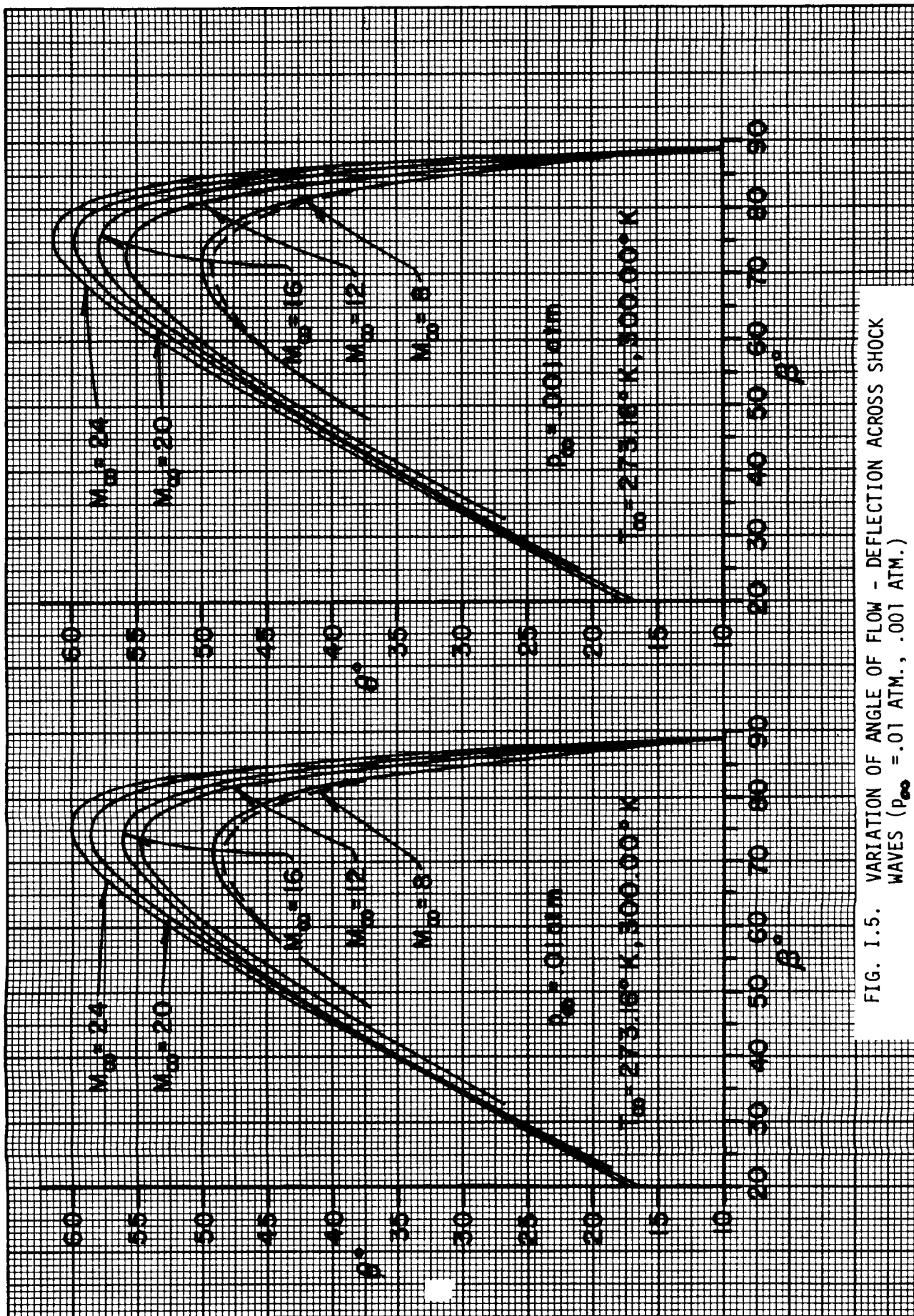


FIG. I.5. VARIATION OF ANGLE OF FLOW - DEFLECTION ACROSS SHOCK WAVES ( $p_\infty = 0.01 \text{ atm.}$ ,  $0.001 \text{ atm.}$ )

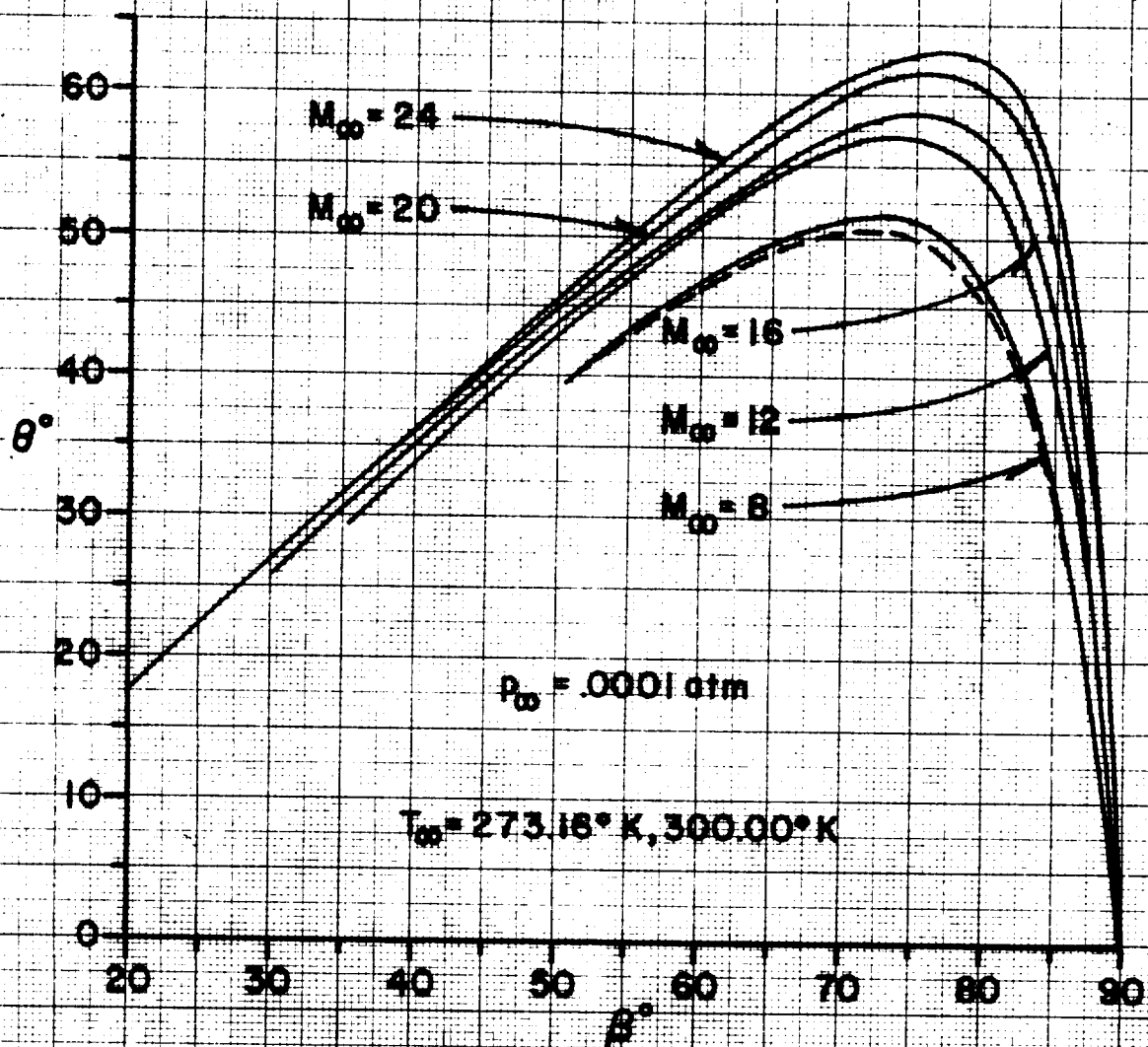


FIG. I.6. VARIATION OF ANGLE OF FLOW - DEFLECTION ACROSS SHOCK WAVES ( $p_\infty = .0001 \text{ ATM.}$ )

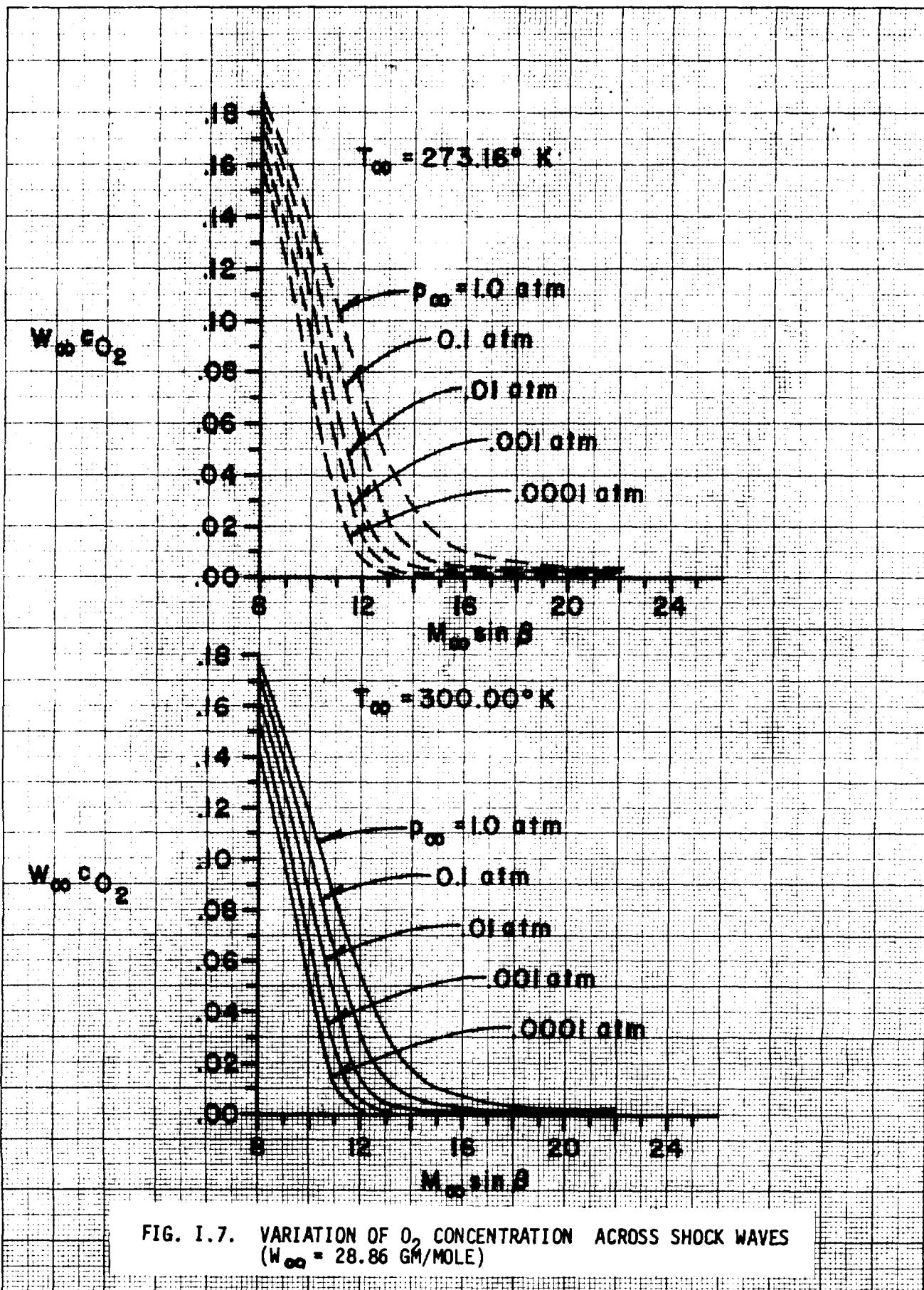


FIG. I.7. VARIATION OF  $\text{O}_2$  CONCENTRATION ACROSS SHOCK WAVES  
( $W_{\infty} = 28.86 \text{ GM/MOLE}$ )



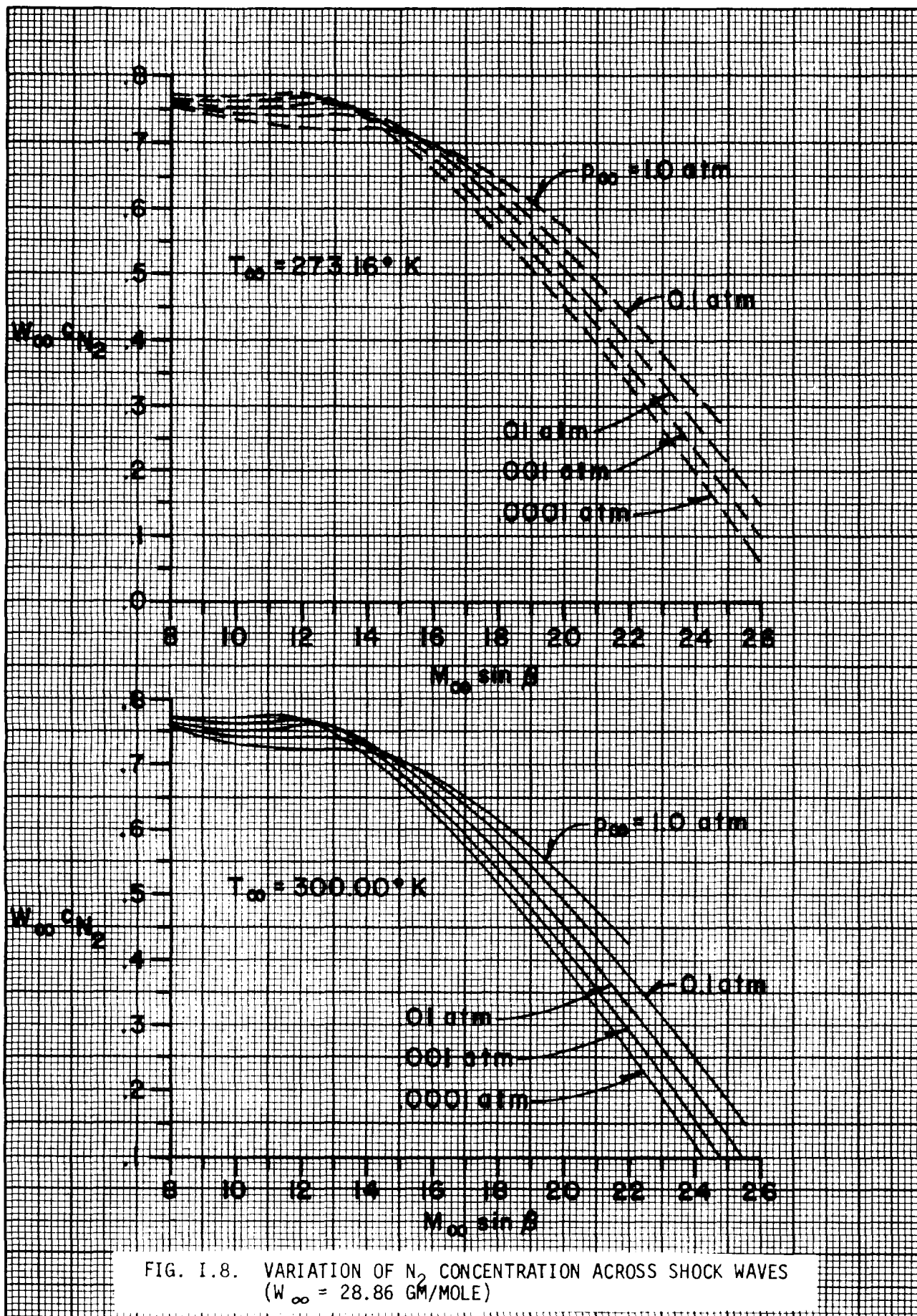


FIG. I.8. VARIATION OF  $N_2$  CONCENTRATION ACROSS SHOCK WAVES  
( $W_{\infty} = 28.86$  GM/MOLE)

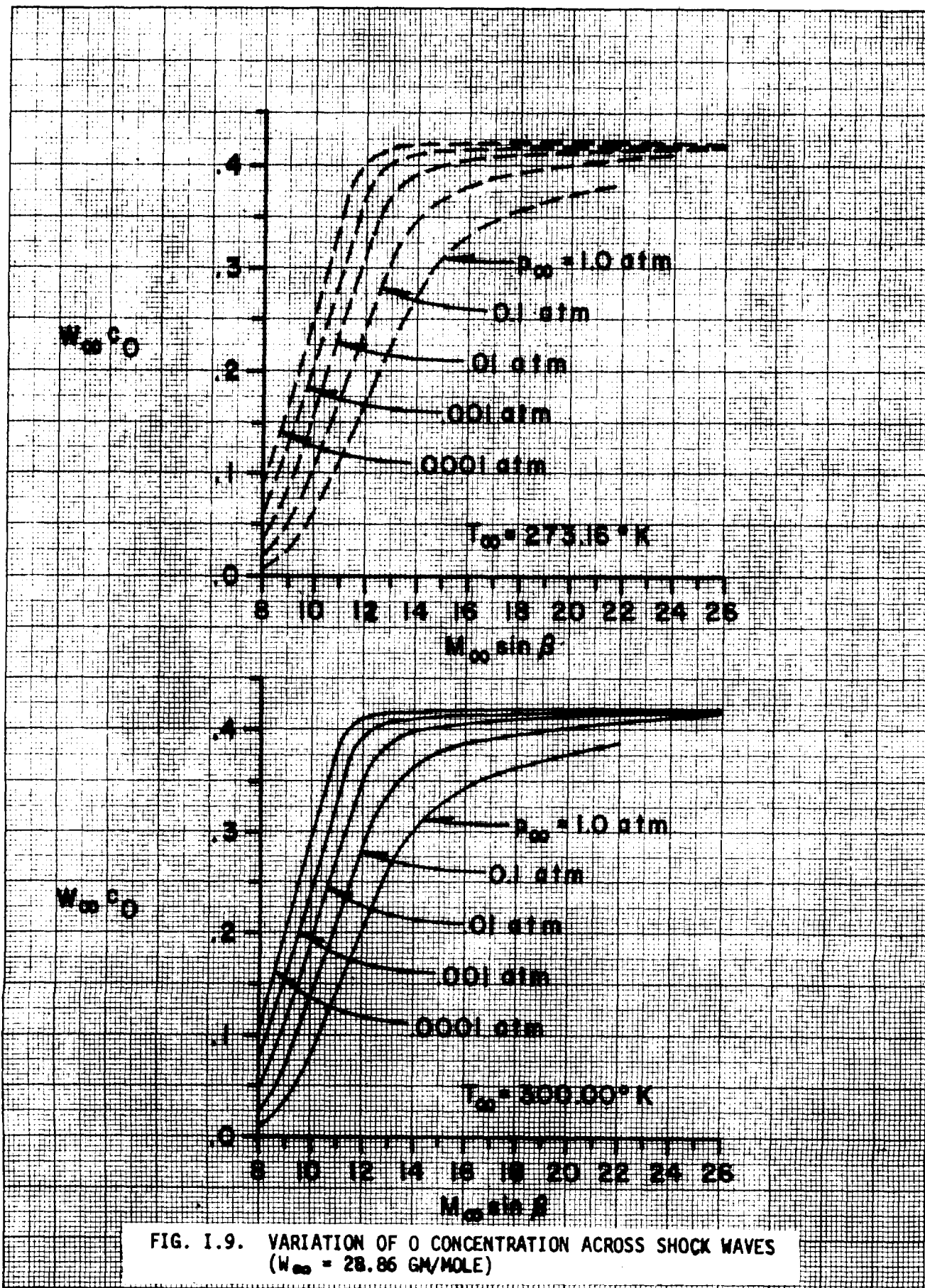


FIG. I.9. VARIATION OF O CONCENTRATION ACROSS SHOCK WAVES  
( $W_{\infty} = 28.86 \text{ GM/MOLE}$ )

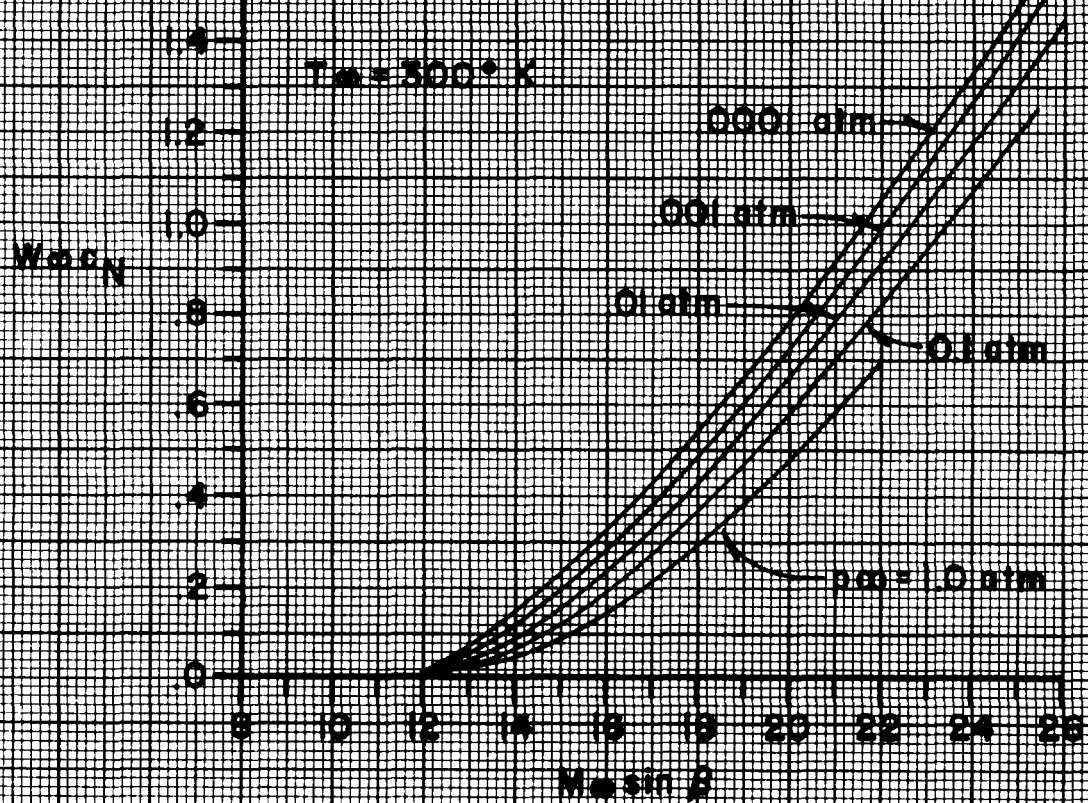
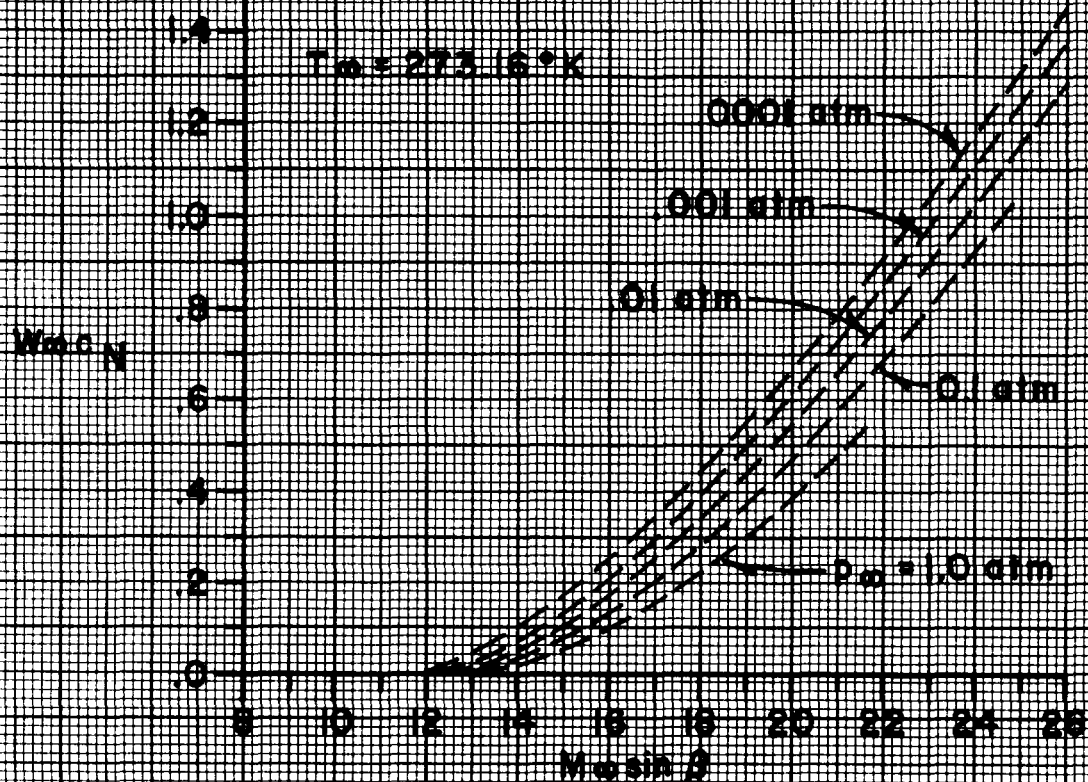
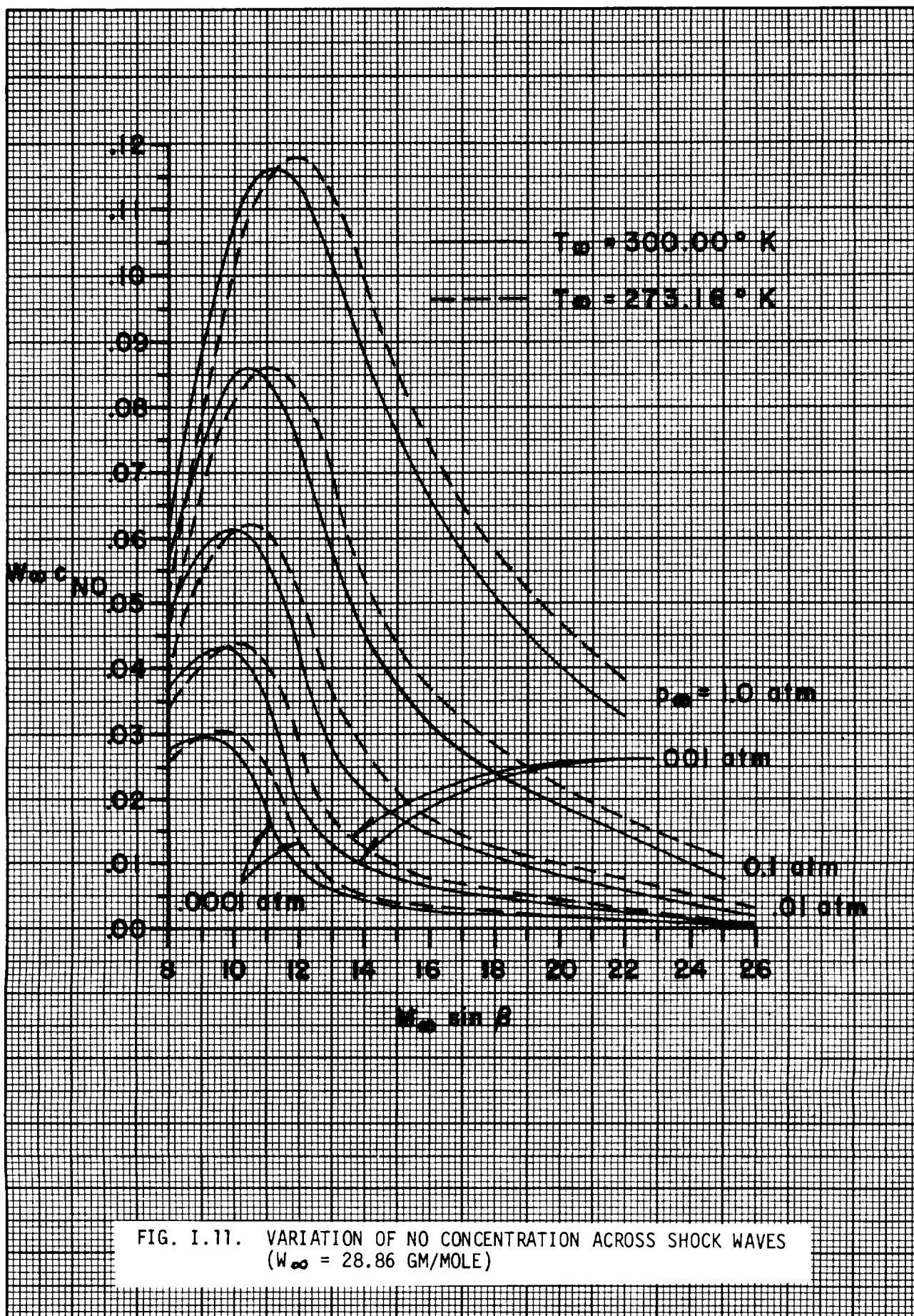


FIG. 1.10. VARIATION OF N CONCENTRATION ACROSS SHOCK WAVES  
 $(W_{\infty} = 28.86 \text{ GM/MOLE})$





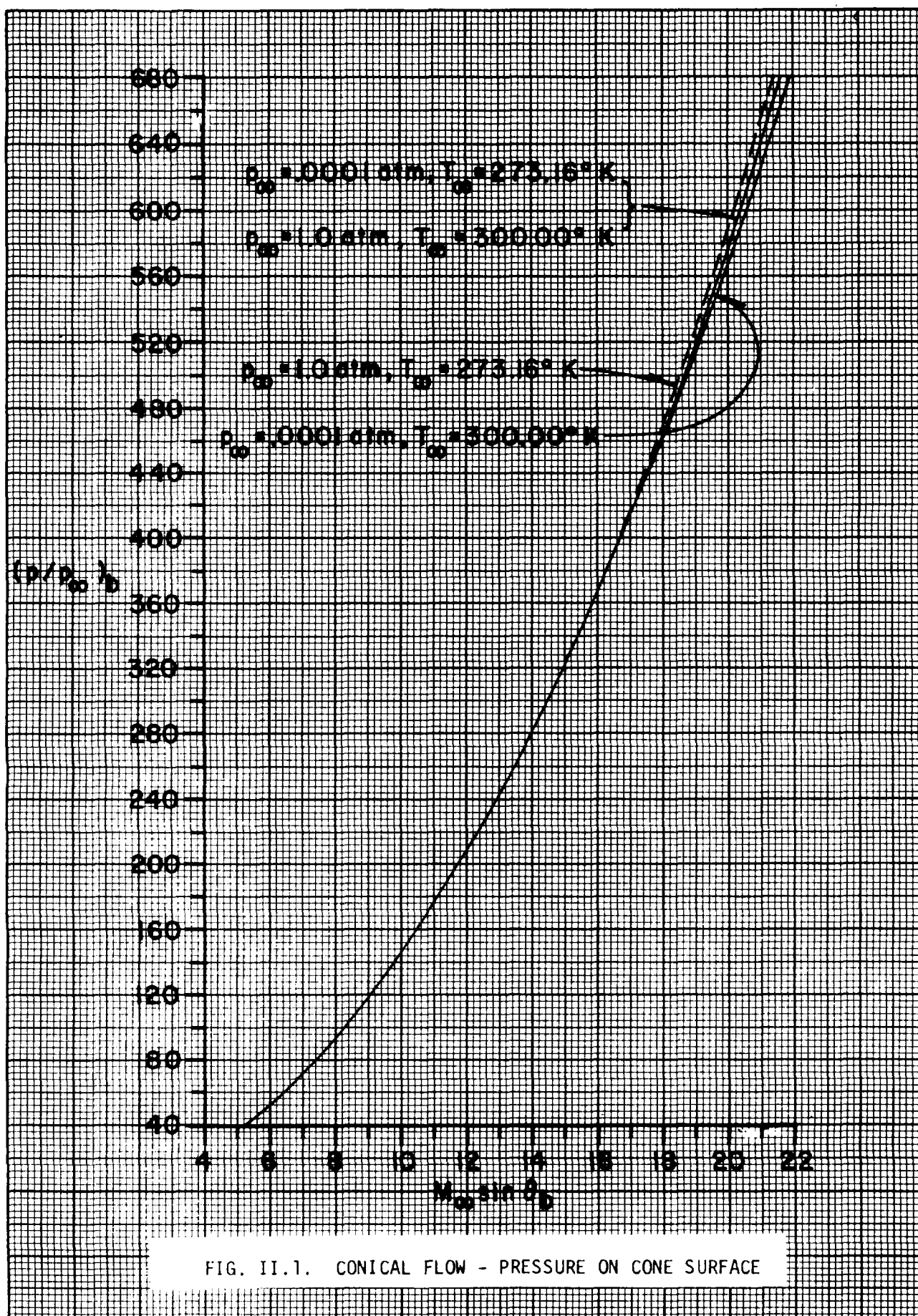


FIG. II.1. CONICAL FLOW - PRESSURE ON CONE SURFACE

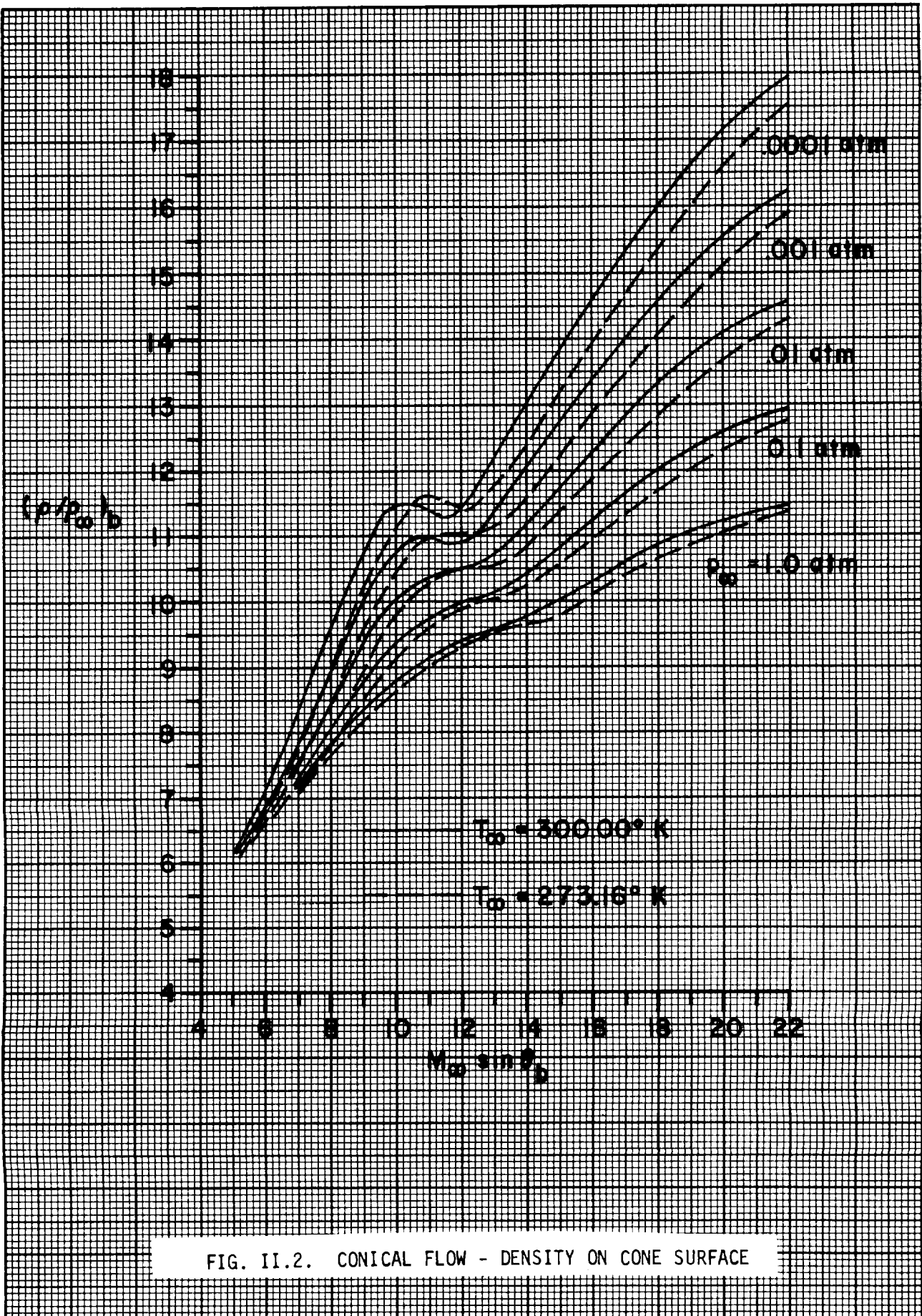


FIG. II.2. CONICAL FLOW - DENSITY ON CONE SURFACE

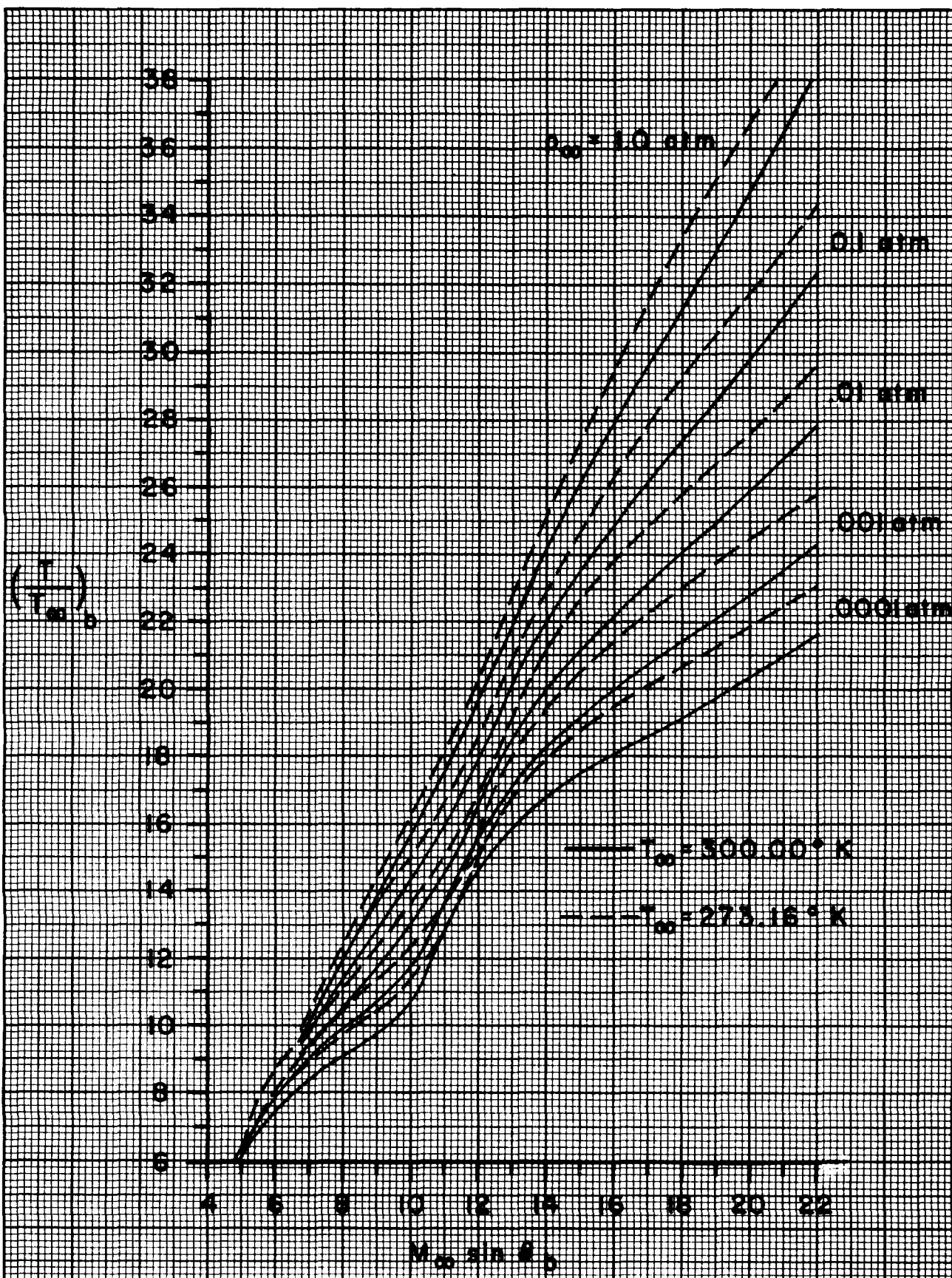


FIG. 11.3. CONICAL FLOW - TEMPERATURE ON CONE SURFACE

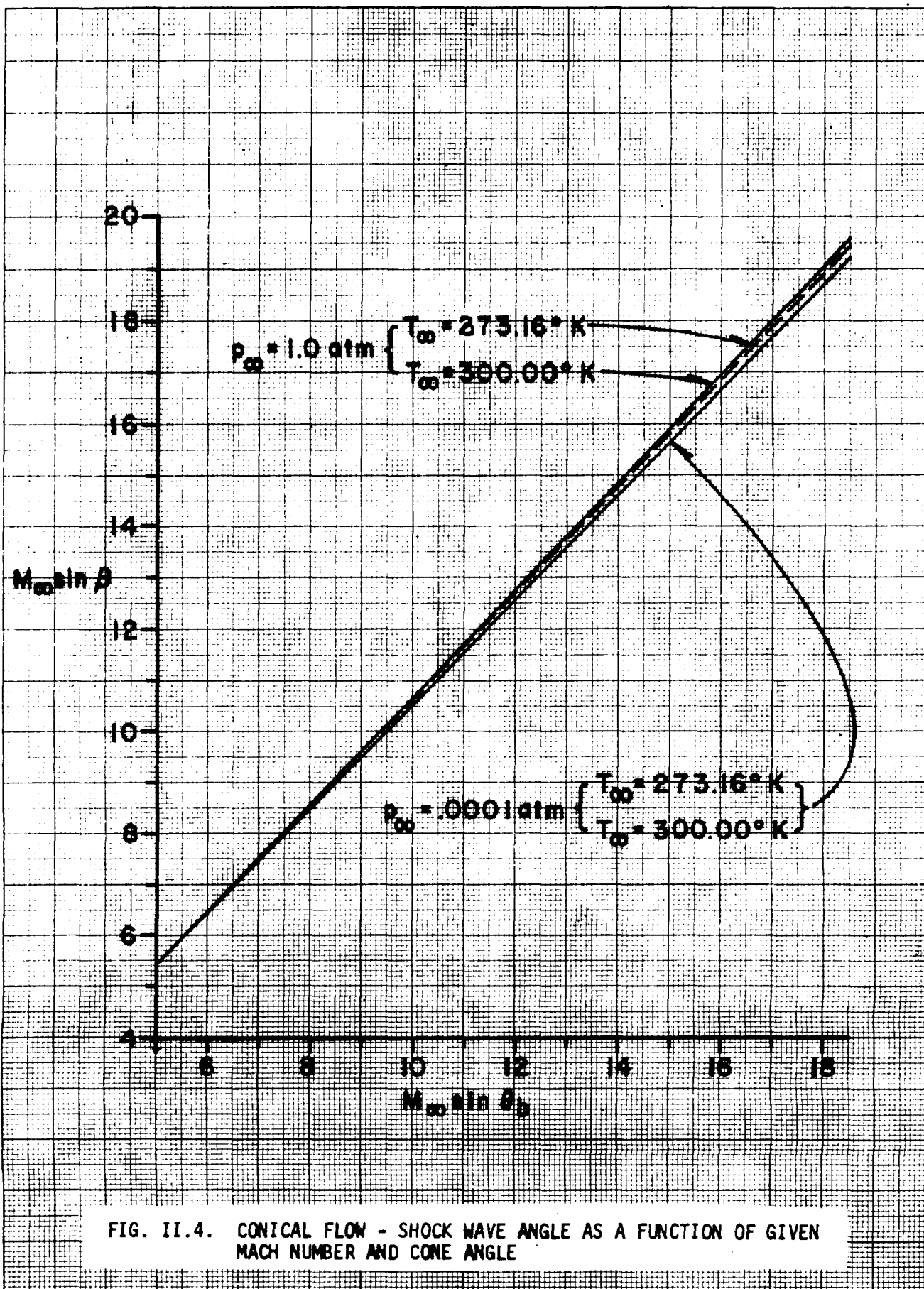


FIG. II.4. CONICAL FLOW - SHOCK WAVE ANGLE AS A FUNCTION OF GIVEN MACH NUMBER AND CONE ANGLE



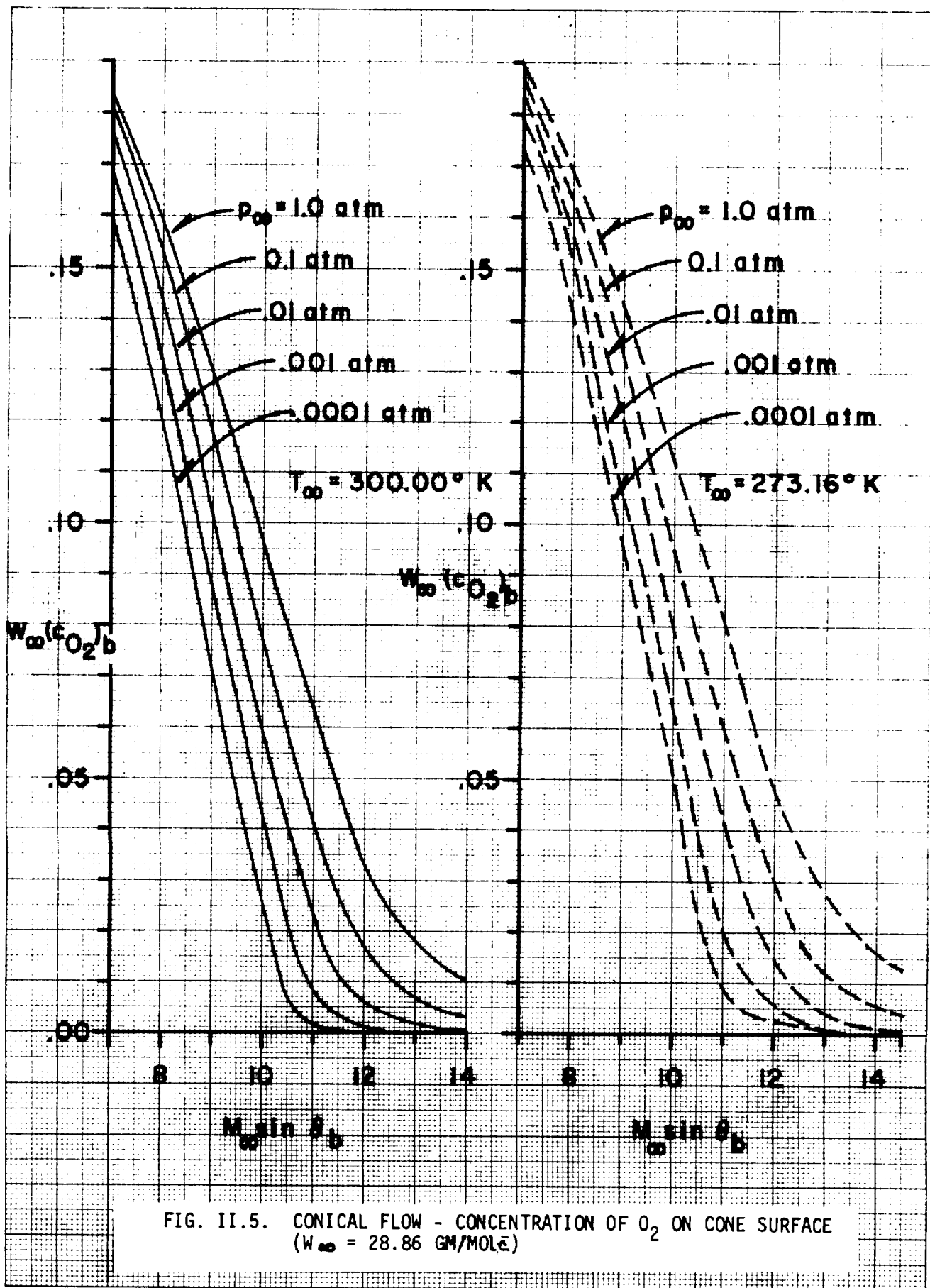


FIG. II.5. CONICAL FLOW - CONCENTRATION OF  $O_2$  ON CONE SURFACE  
( $W_\infty = 28.86 \text{ GM/MOL } \bar{c}$ )

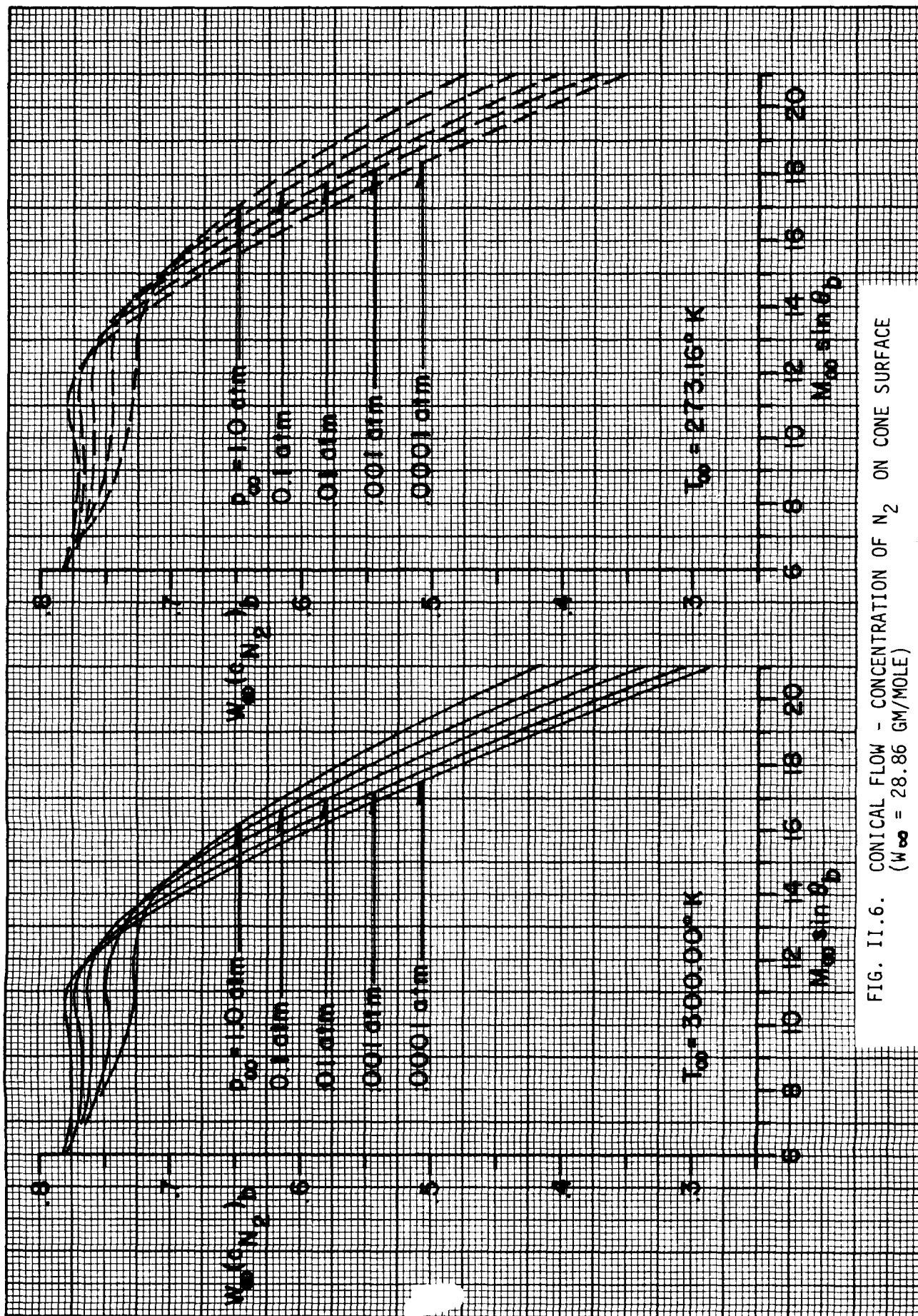
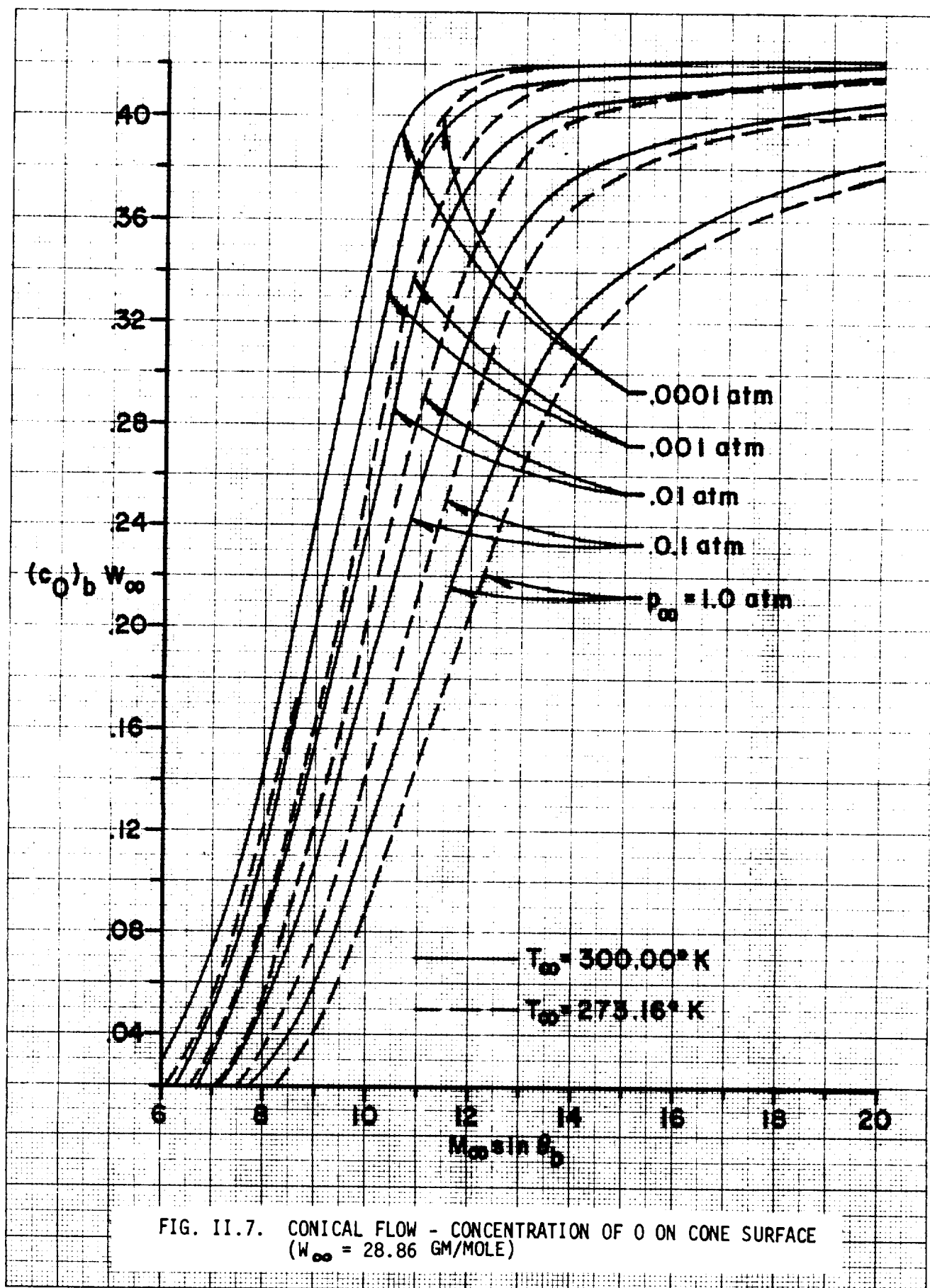


FIG. II.6. CONICAL FLOW - CONCENTRATION OF  $N_2$  ON CONE SURFACE  
( $W_\infty = 28.86$  GM/MOLE)



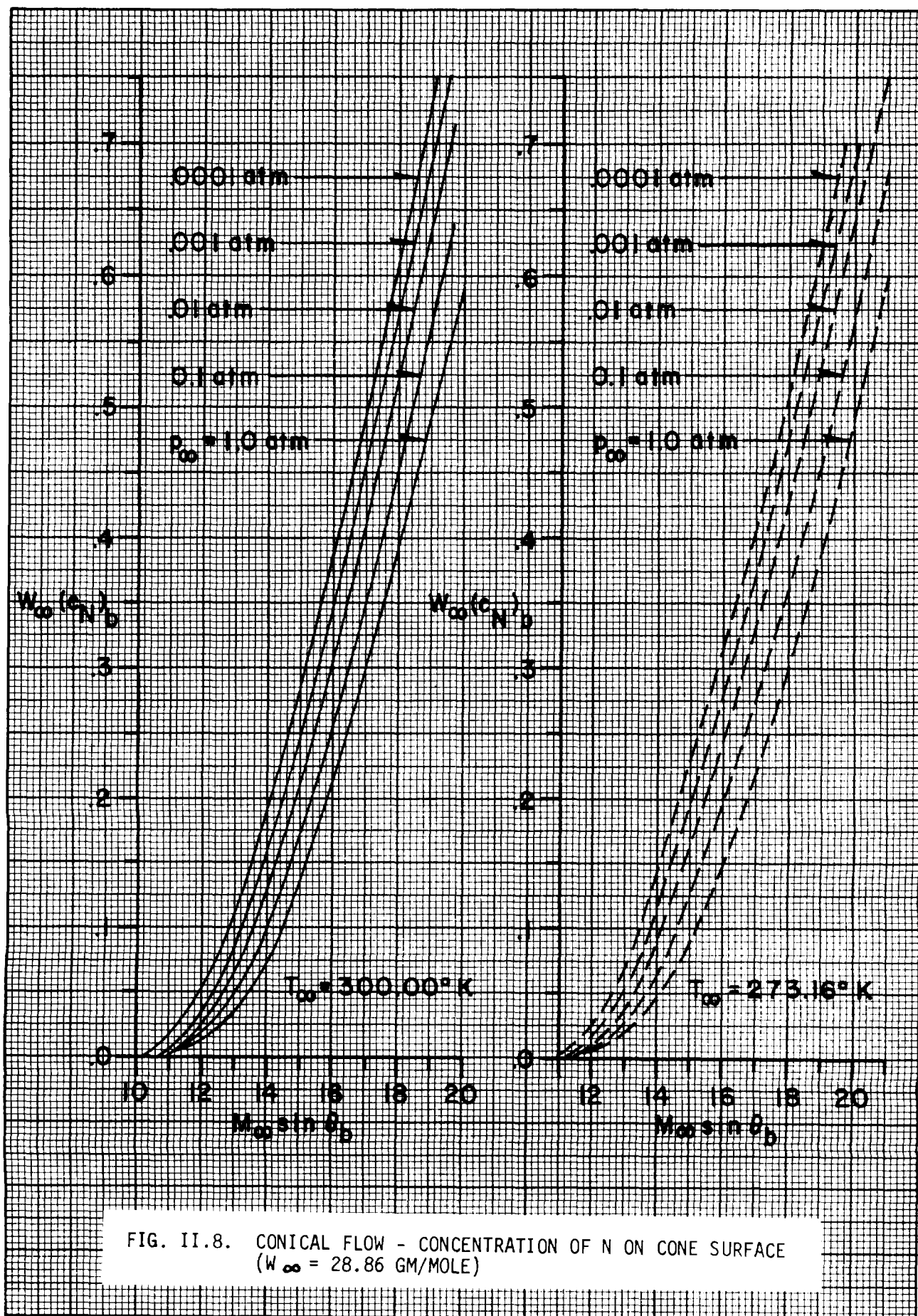
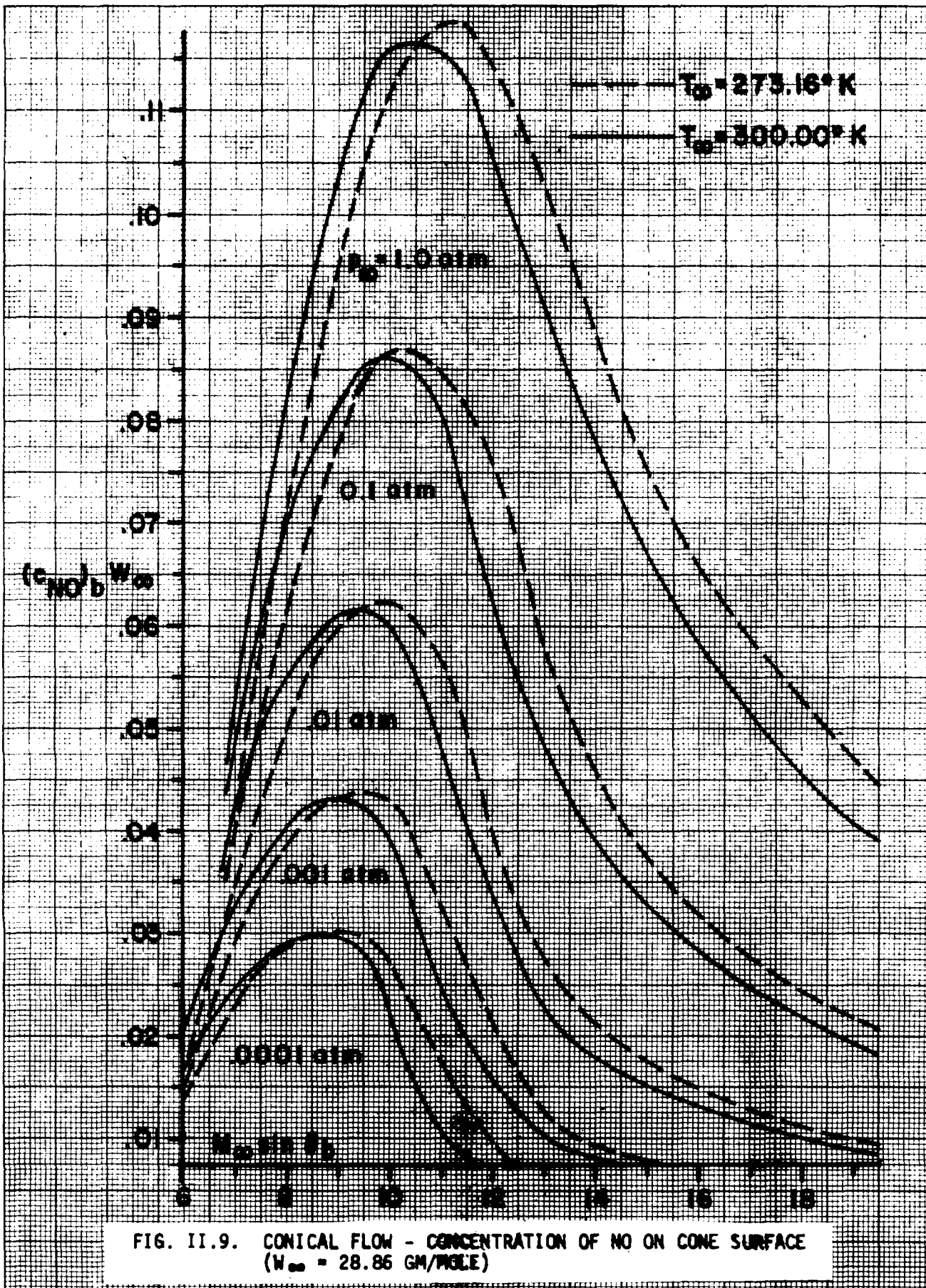


FIG. II.8. CONICAL FLOW - CONCENTRATION OF N ON CONE SURFACE  
( $W_{\infty} = 28.86$  GM/MOLE)





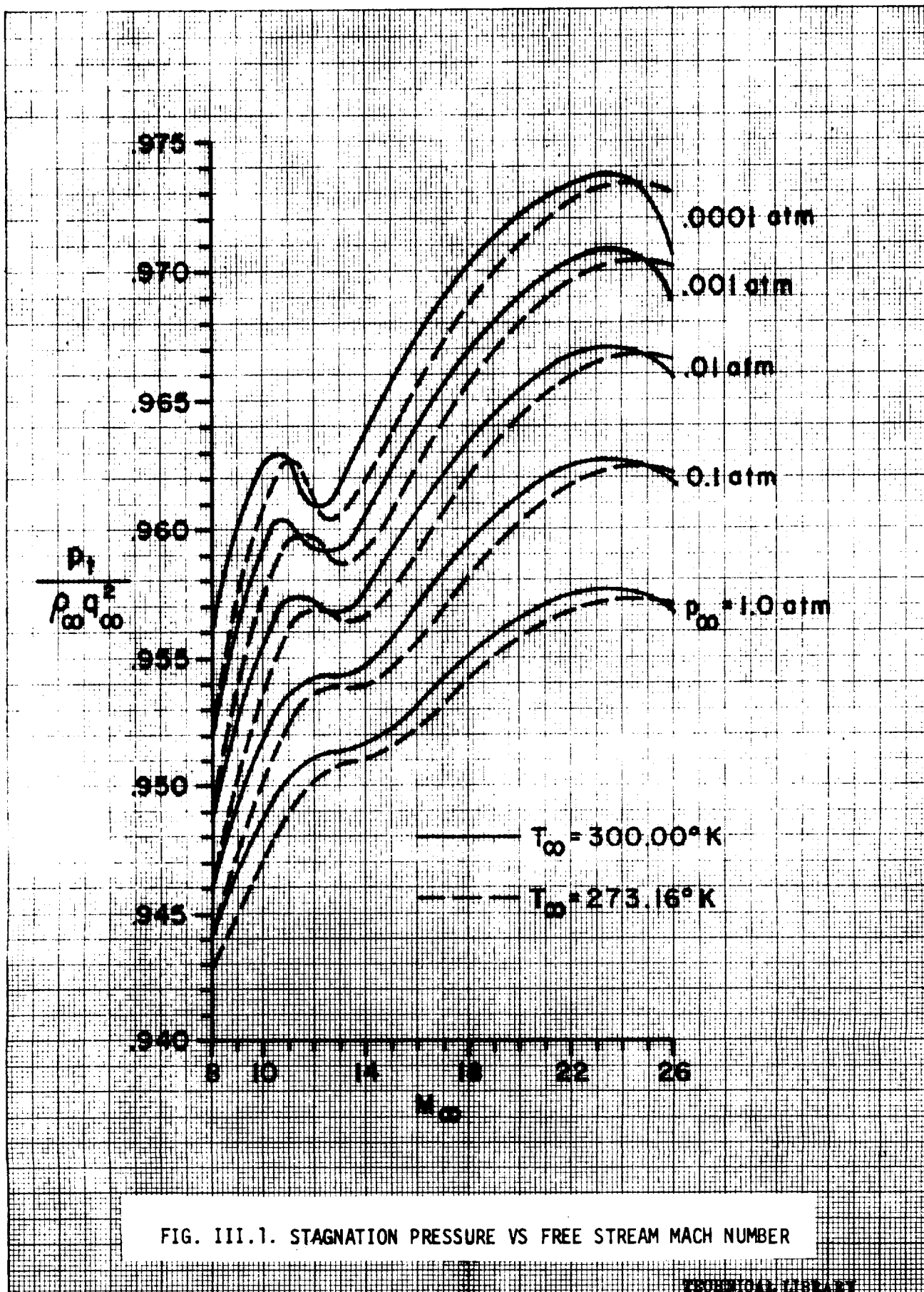


FIG. III.1. STAGNATION PRESSURE VS FREE STREAM MACH NUMBER

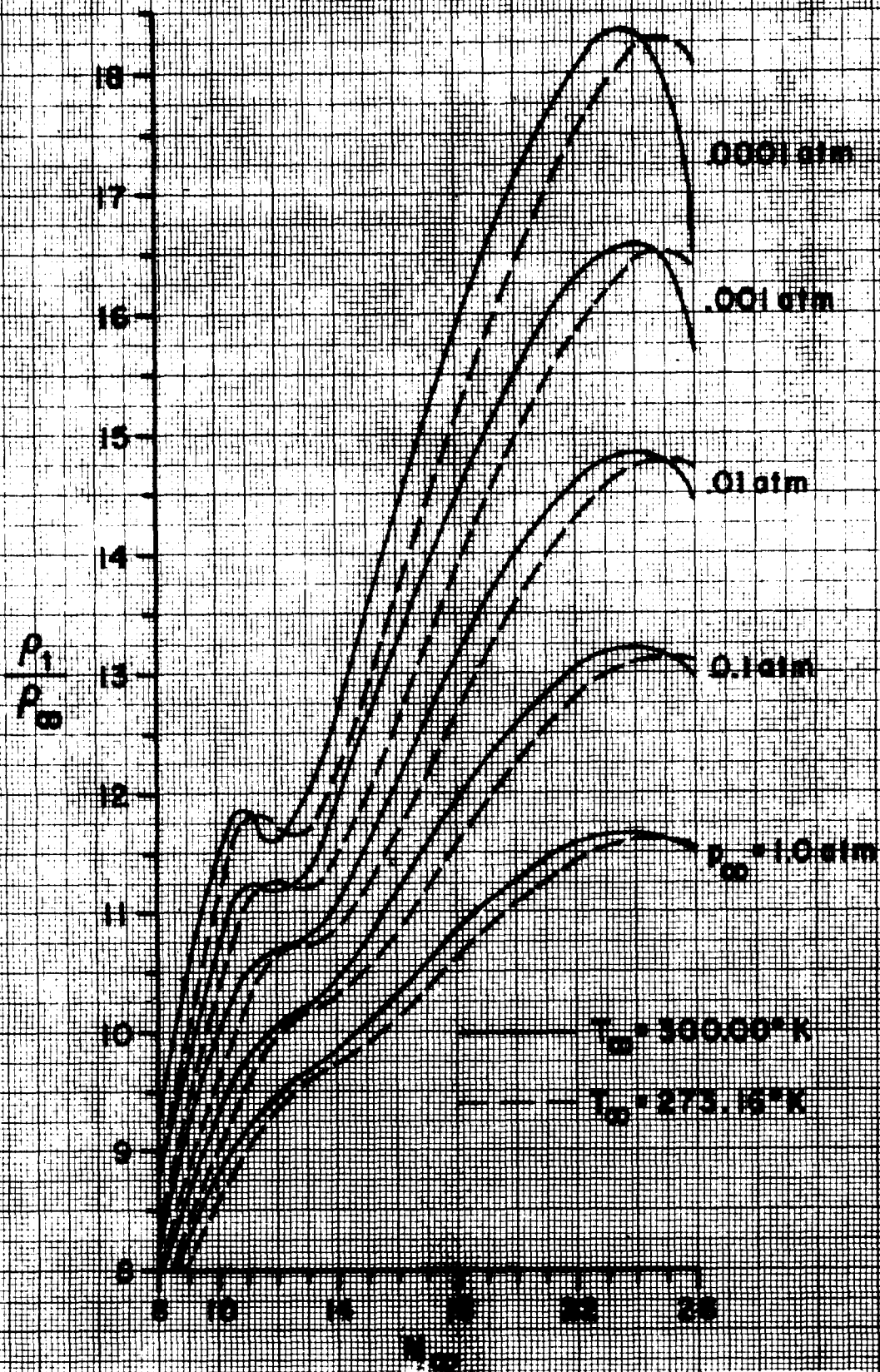


FIG. III.2. STAGNATION DENSITY VS FREE STREAM MACH NUMBER

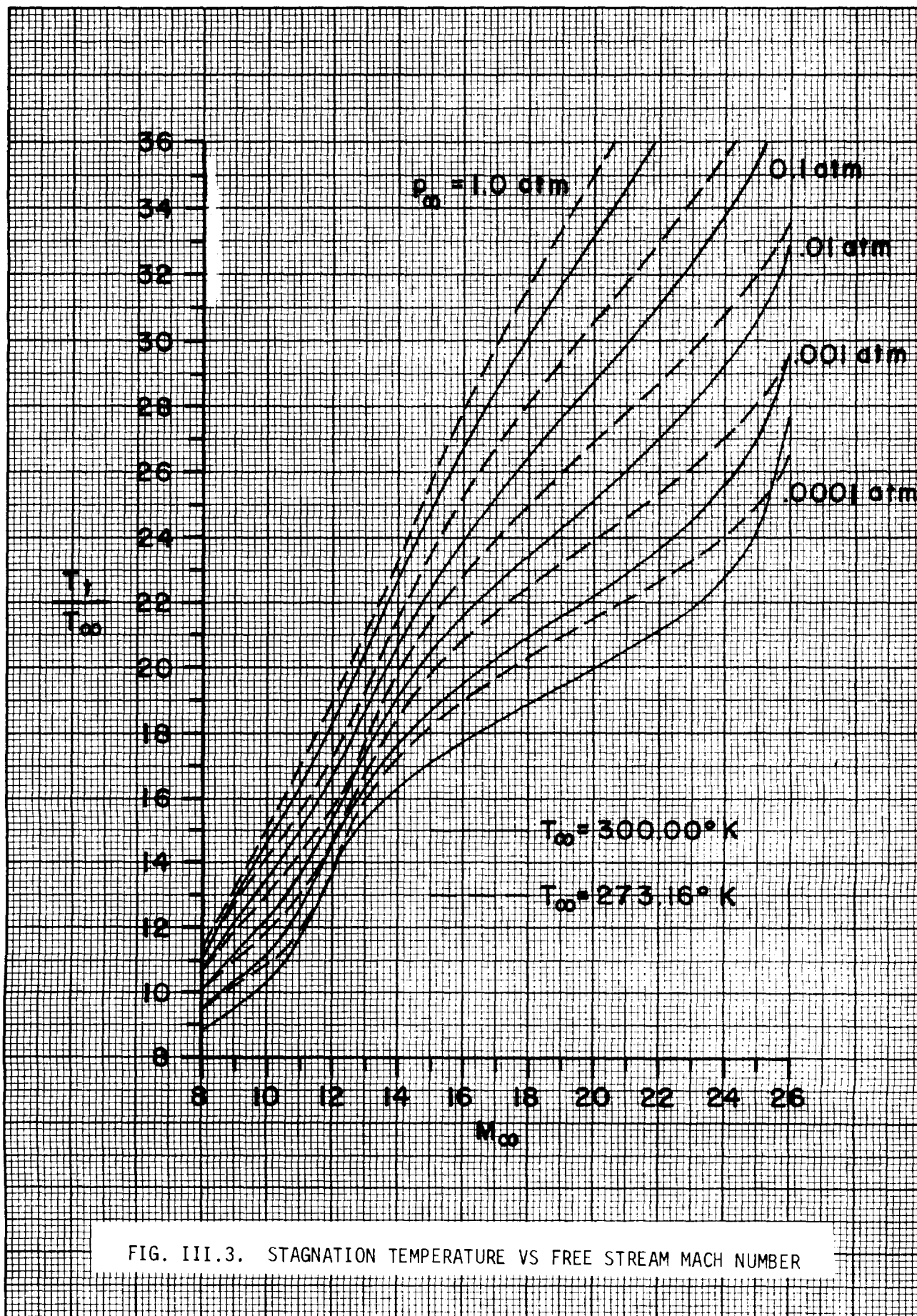


FIG. III.3. STAGNATION TEMPERATURE VS FREE STREAM MACH NUMBER



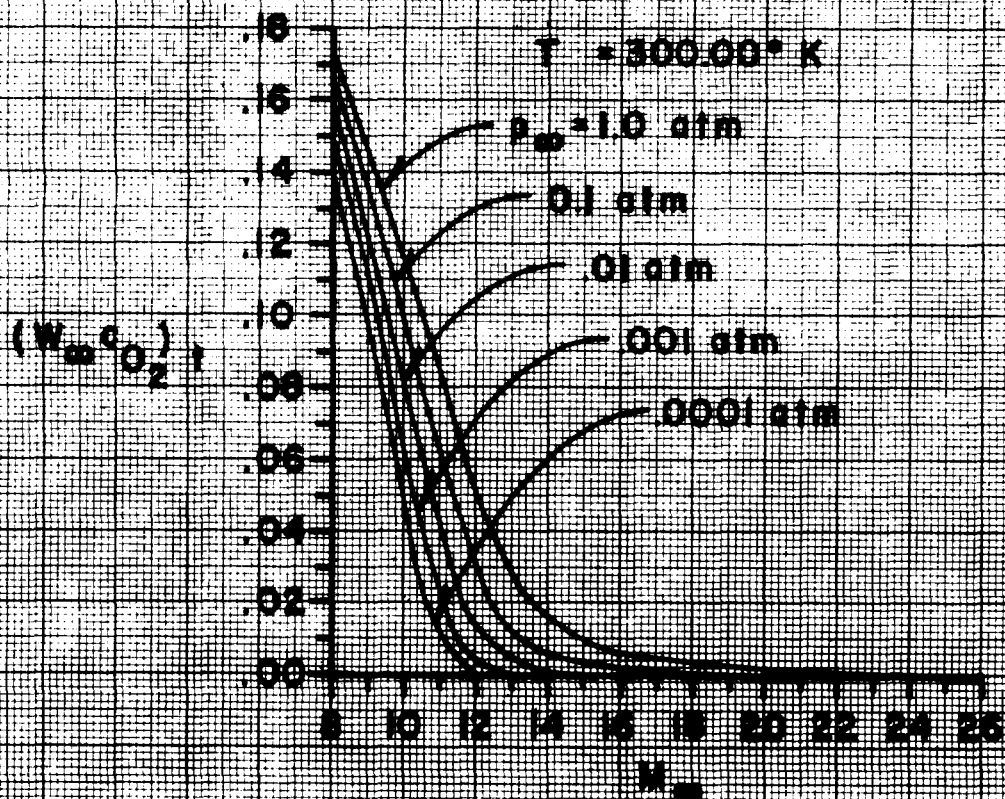
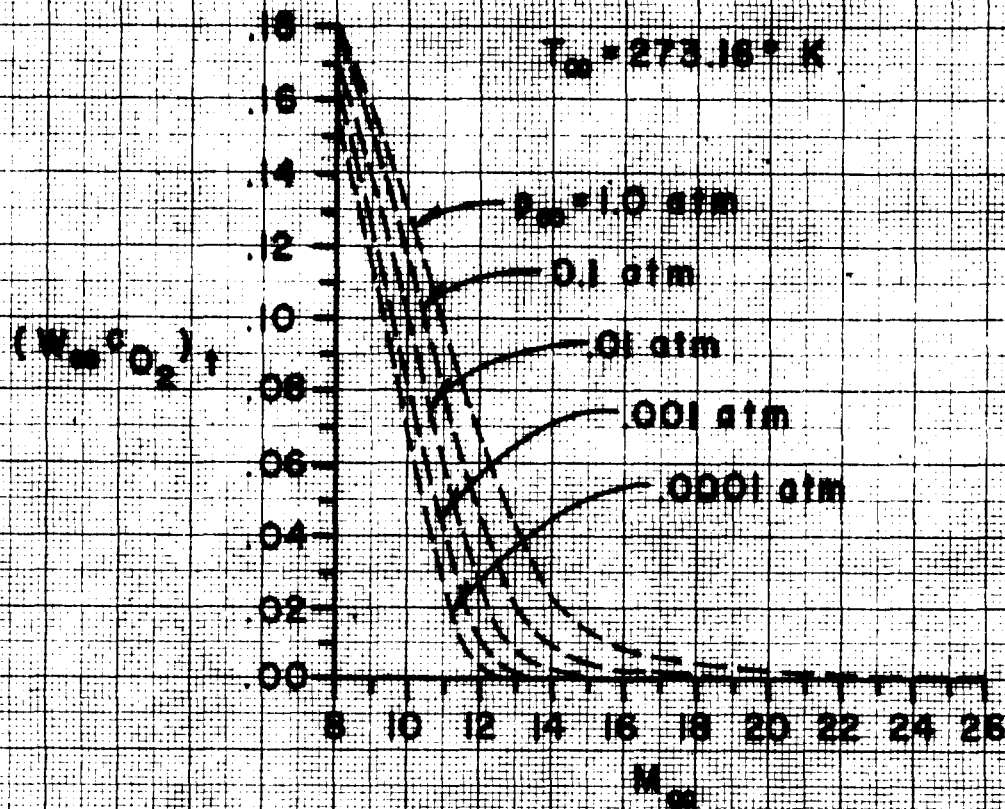


FIG. III.4. STAGNATION  $O_2$  CONCENTRATION VS FREE STREAM MACH NUMBER  
 $(W_{\infty} = 28.86 \text{ GM/MOLE})$

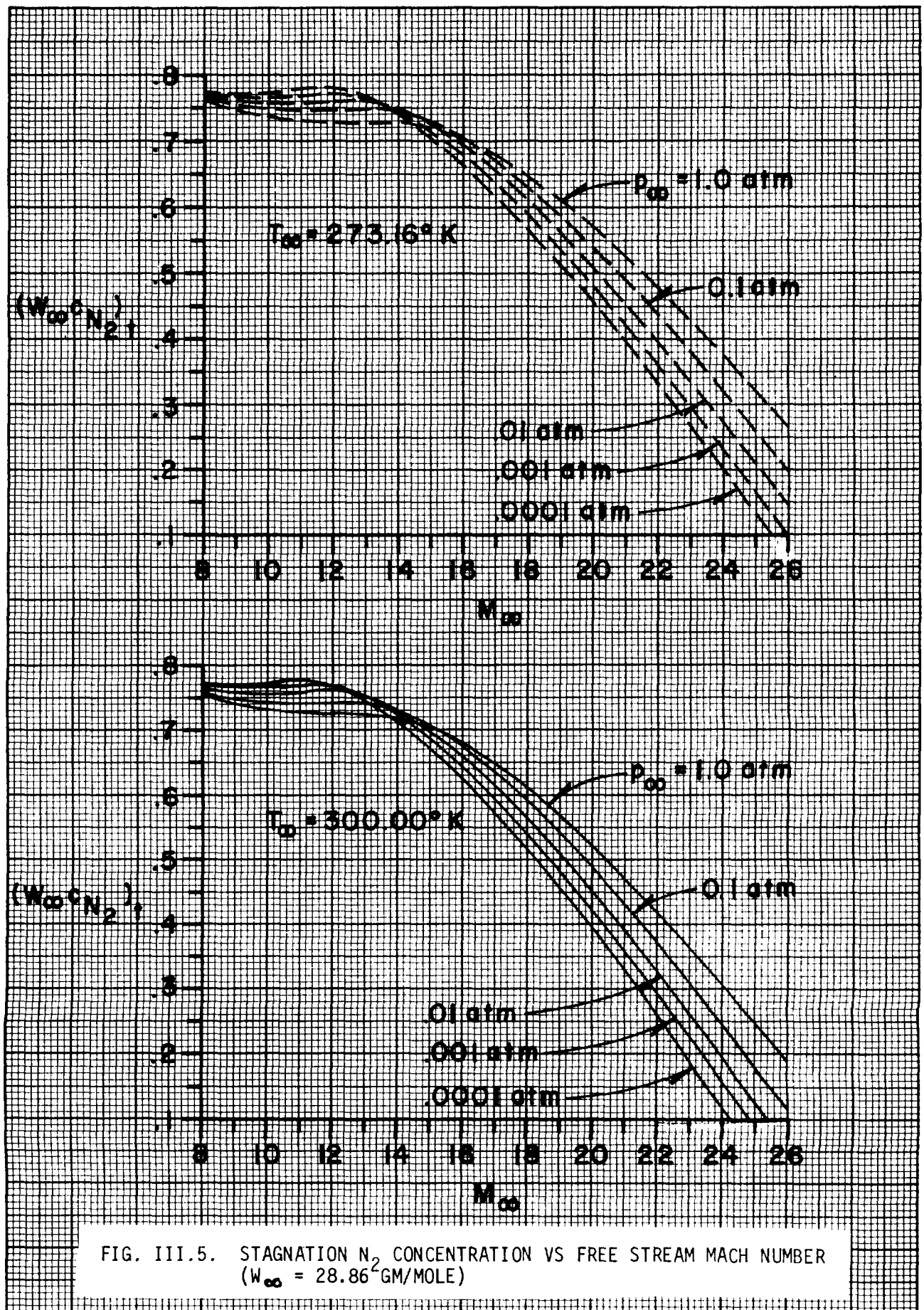


FIG. III.5. STAGNATION  $N_2$  CONCENTRATION VS FREE STREAM MACH NUMBER  
 $(W_\infty = 28.86 \text{ GM/MOLE})$

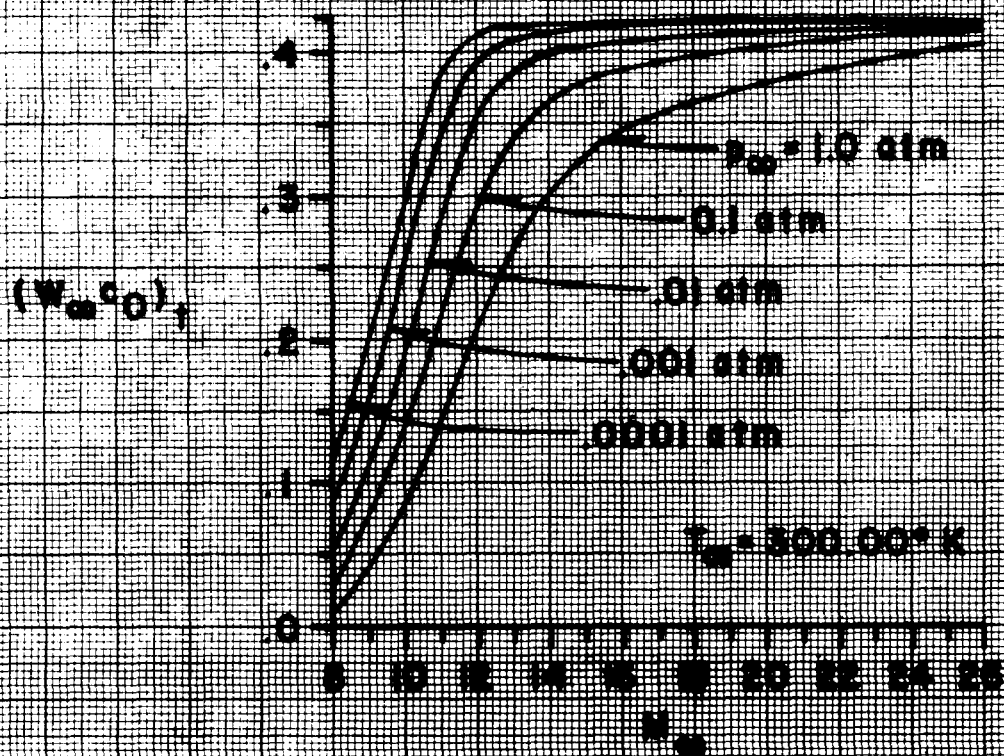
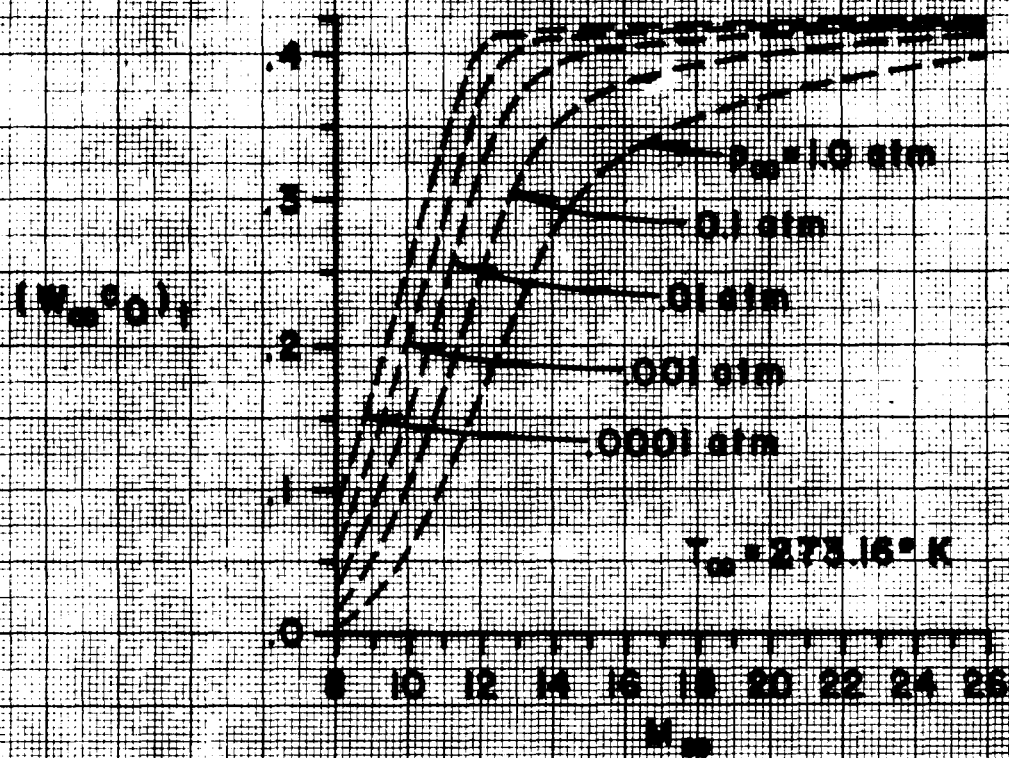


FIG. III.6. STAGNATION O CONCENTRATION VS FREE STREAM MACH NUMBER  
( $W_{O_2} = 28.86 \text{ GM/MOLE}$ )

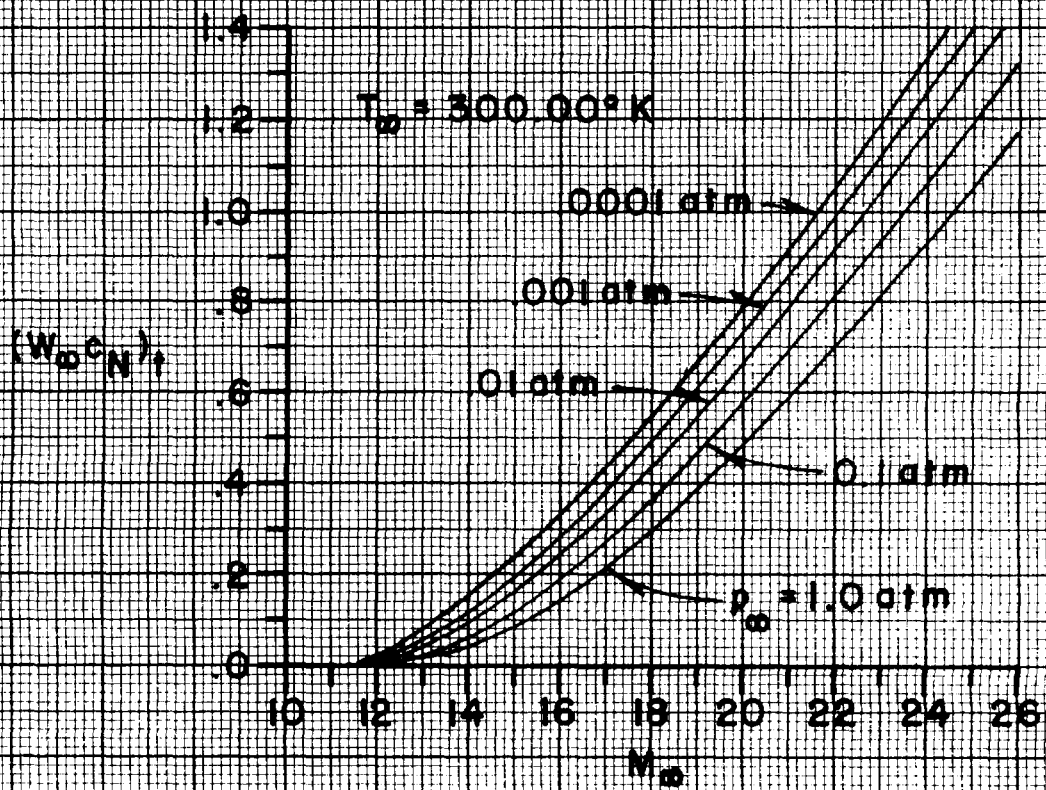
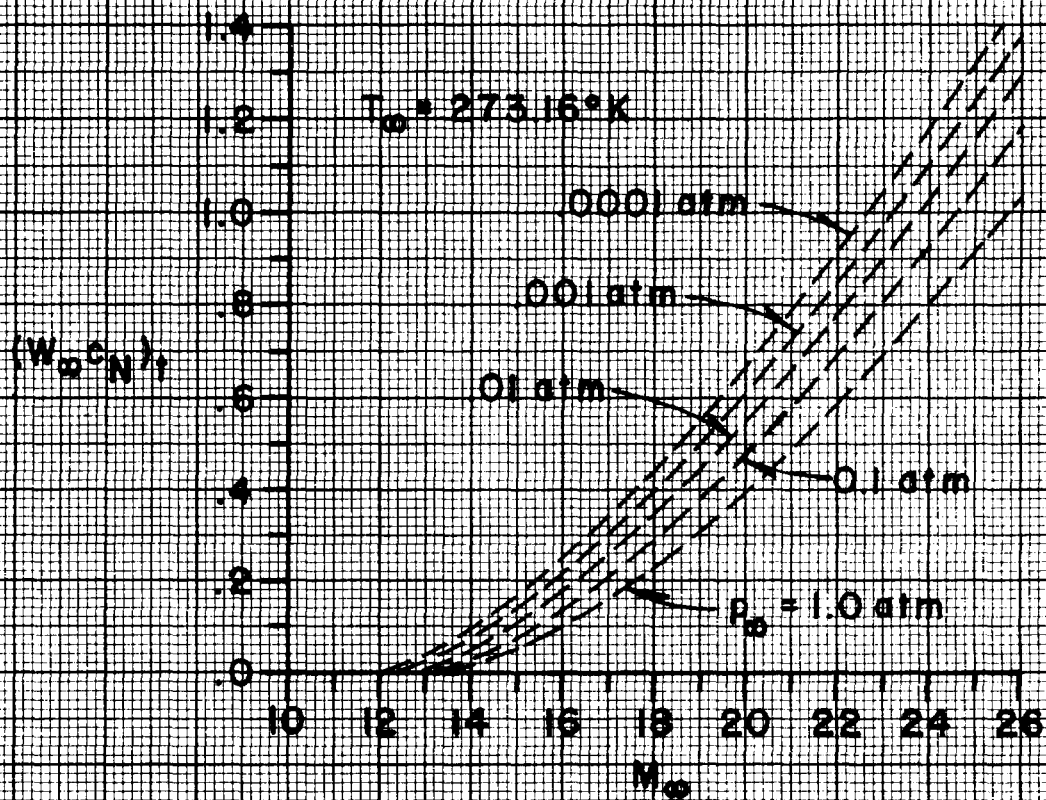
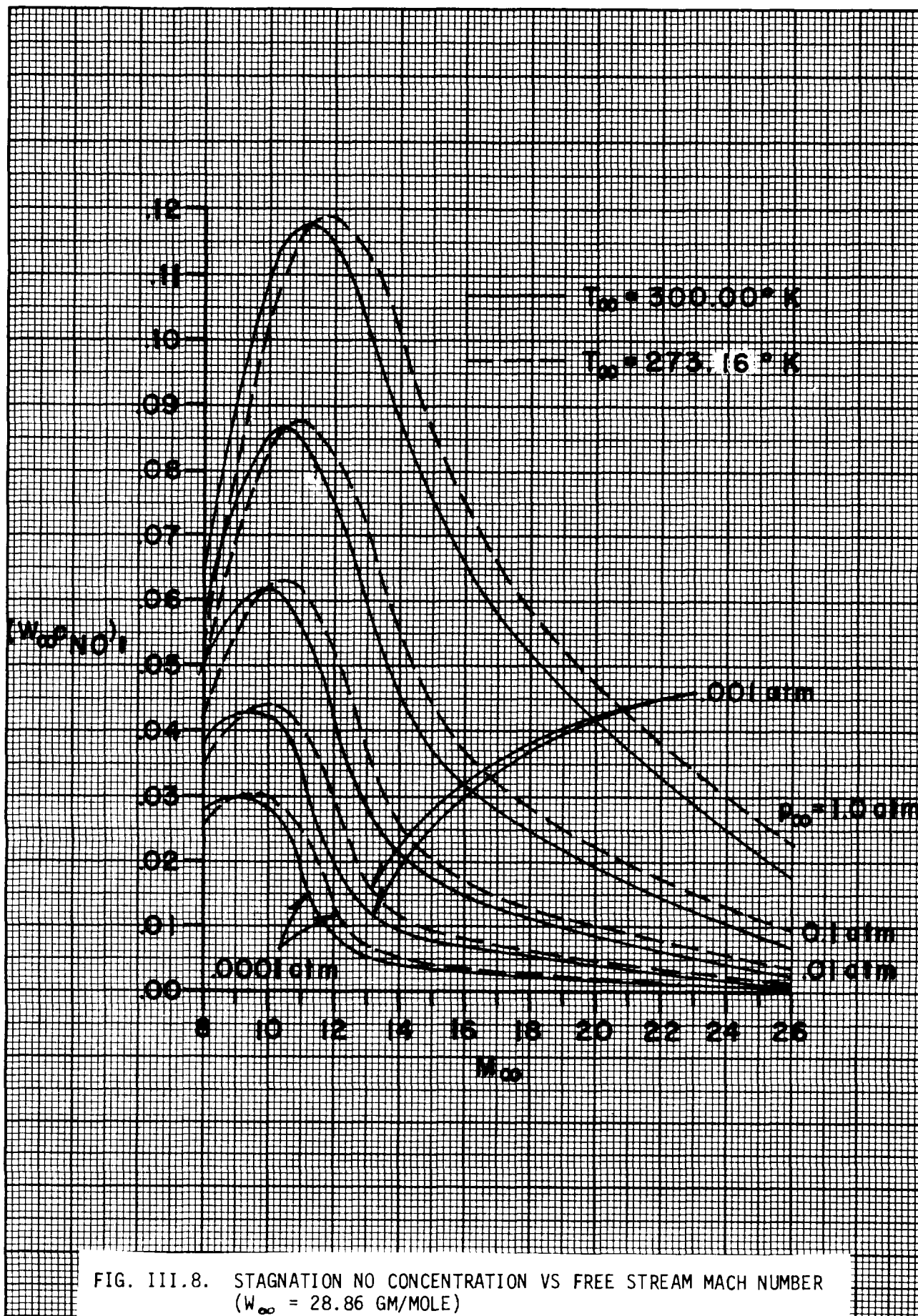


FIG. III.7. STAGNATION N CONCENTRATION VS FREE STREAM MACH NUMBER  
( $W_{\infty} = 28.86 \text{ GM/MOLE}$ )





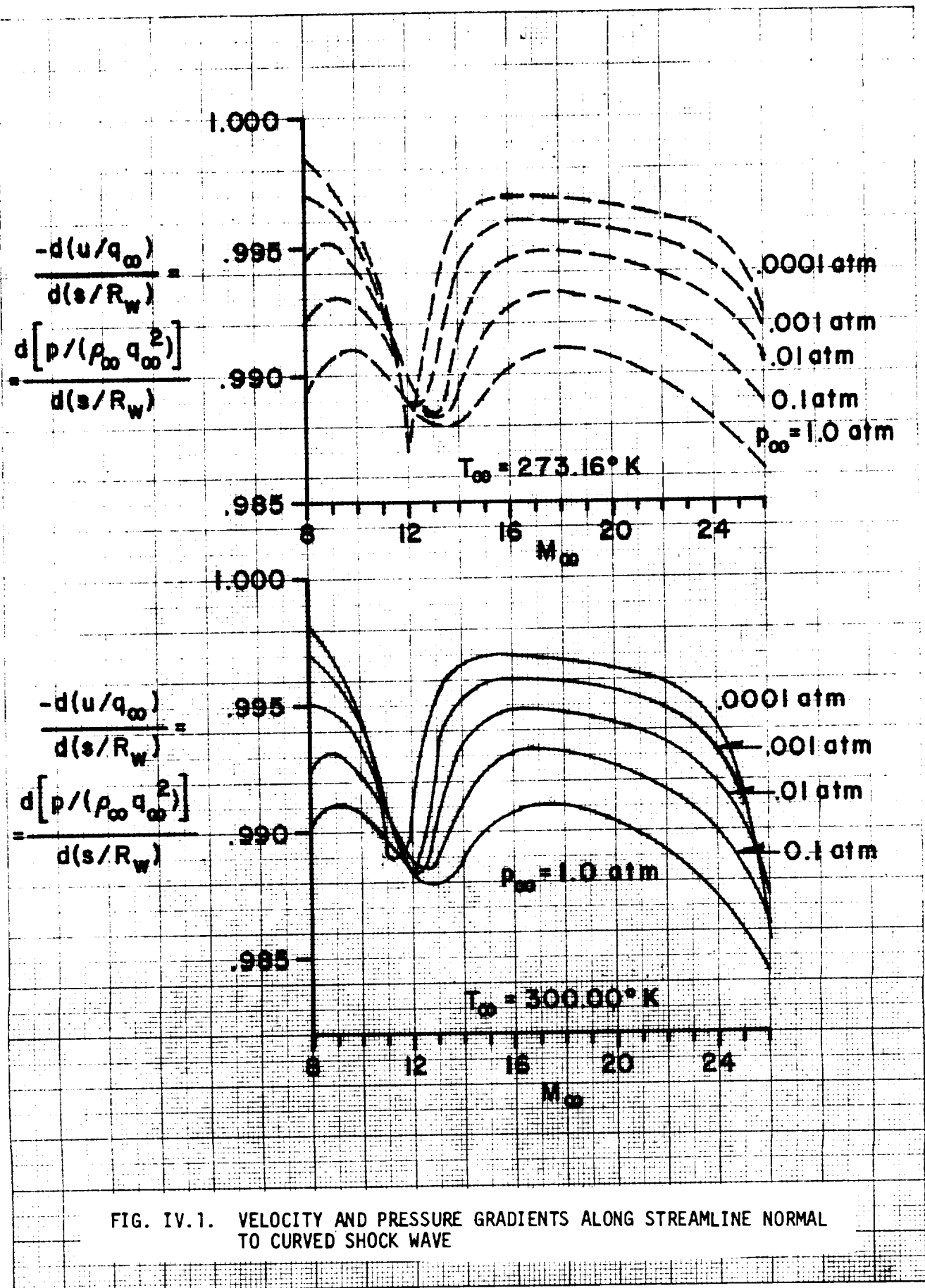


FIG. IV.1. VELOCITY AND PRESSURE GRADIENTS ALONG STREAMLINE NORMAL TO CURVED SHOCK WAVE

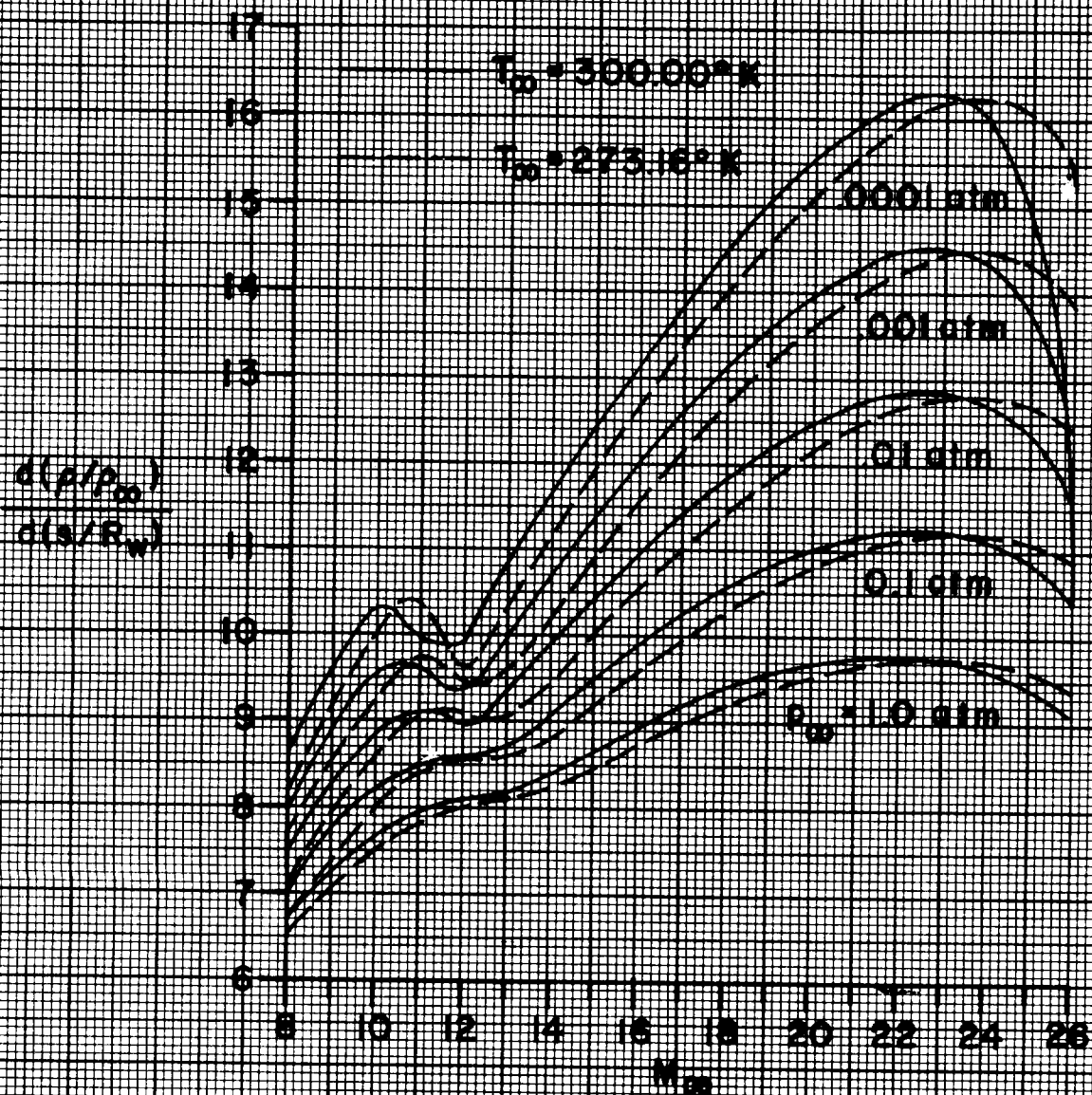


FIG. IV.2. DENSITY GRADIENT ALONG STREAMLINE NORMAL TO CURVED SHOCK WAVE

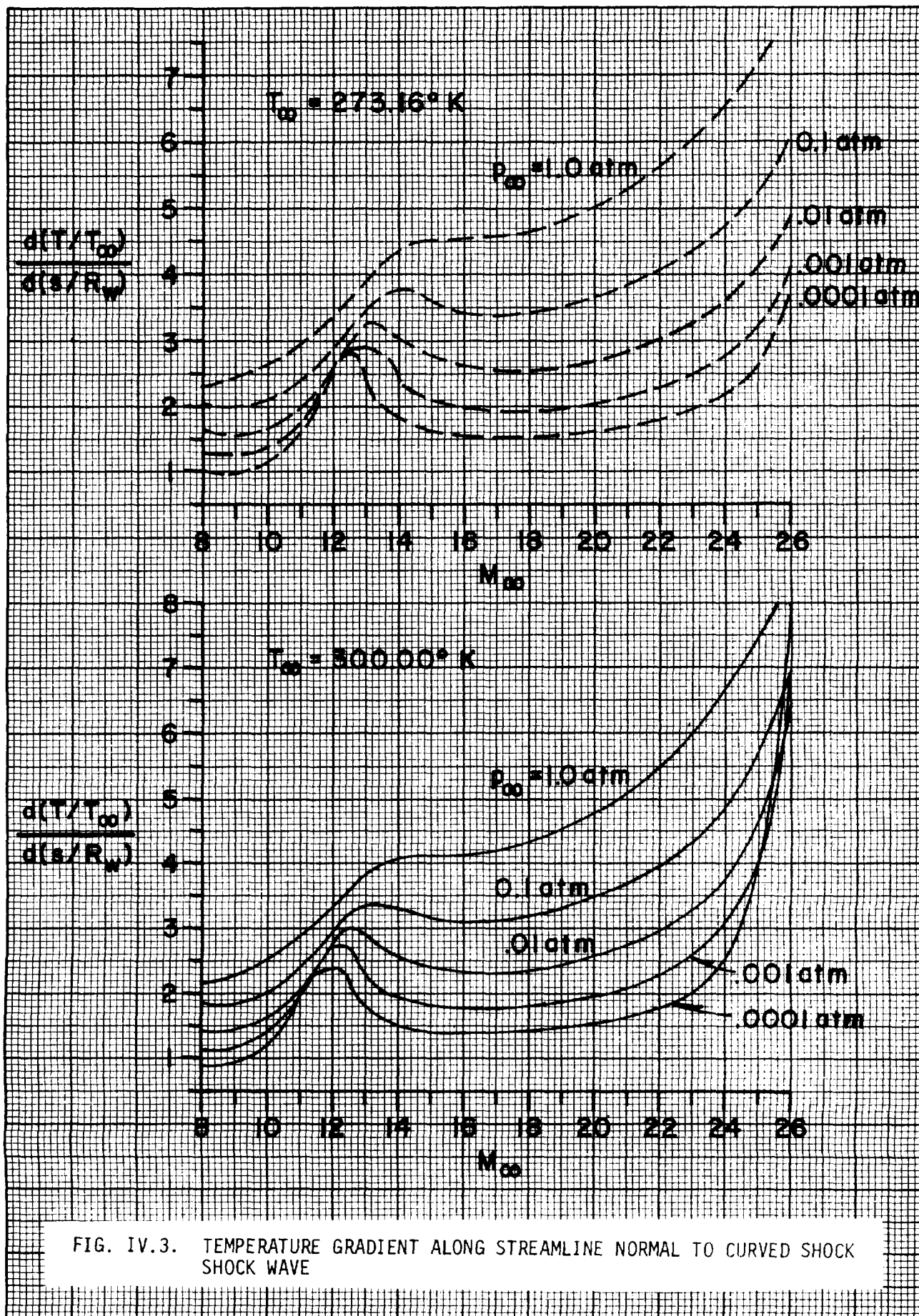


FIG. IV.3. TEMPERATURE GRADIENT ALONG STREAMLINE NORMAL TO CURVED SHOCK SHOCK WAVE



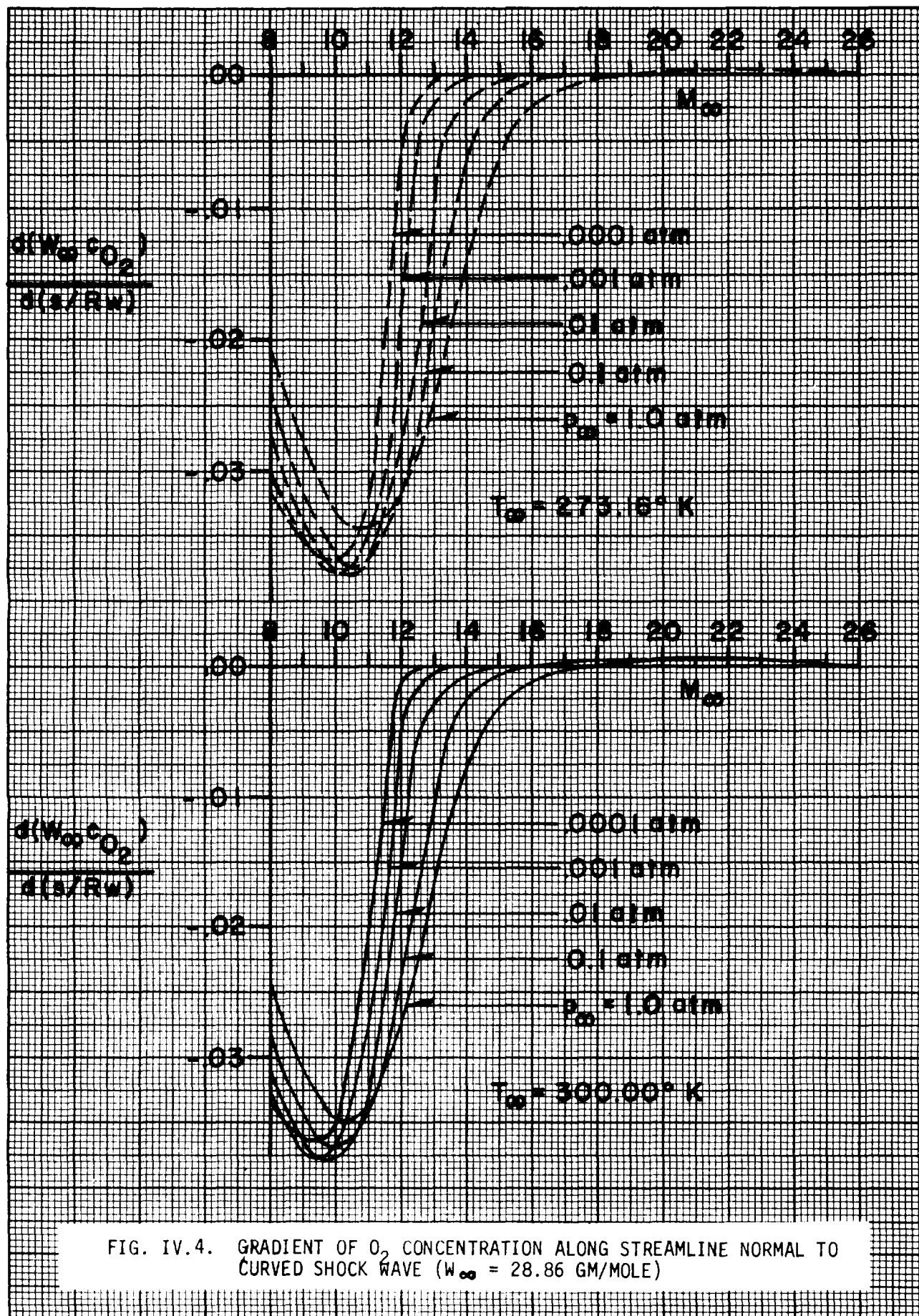


FIG. IV.4. GRADIENT OF  $O_2$  CONCENTRATION ALONG STREAMLINE NORMAL TO CURVED SHOCK WAVE ( $w_\infty = 28.86$  GM/MOLE)

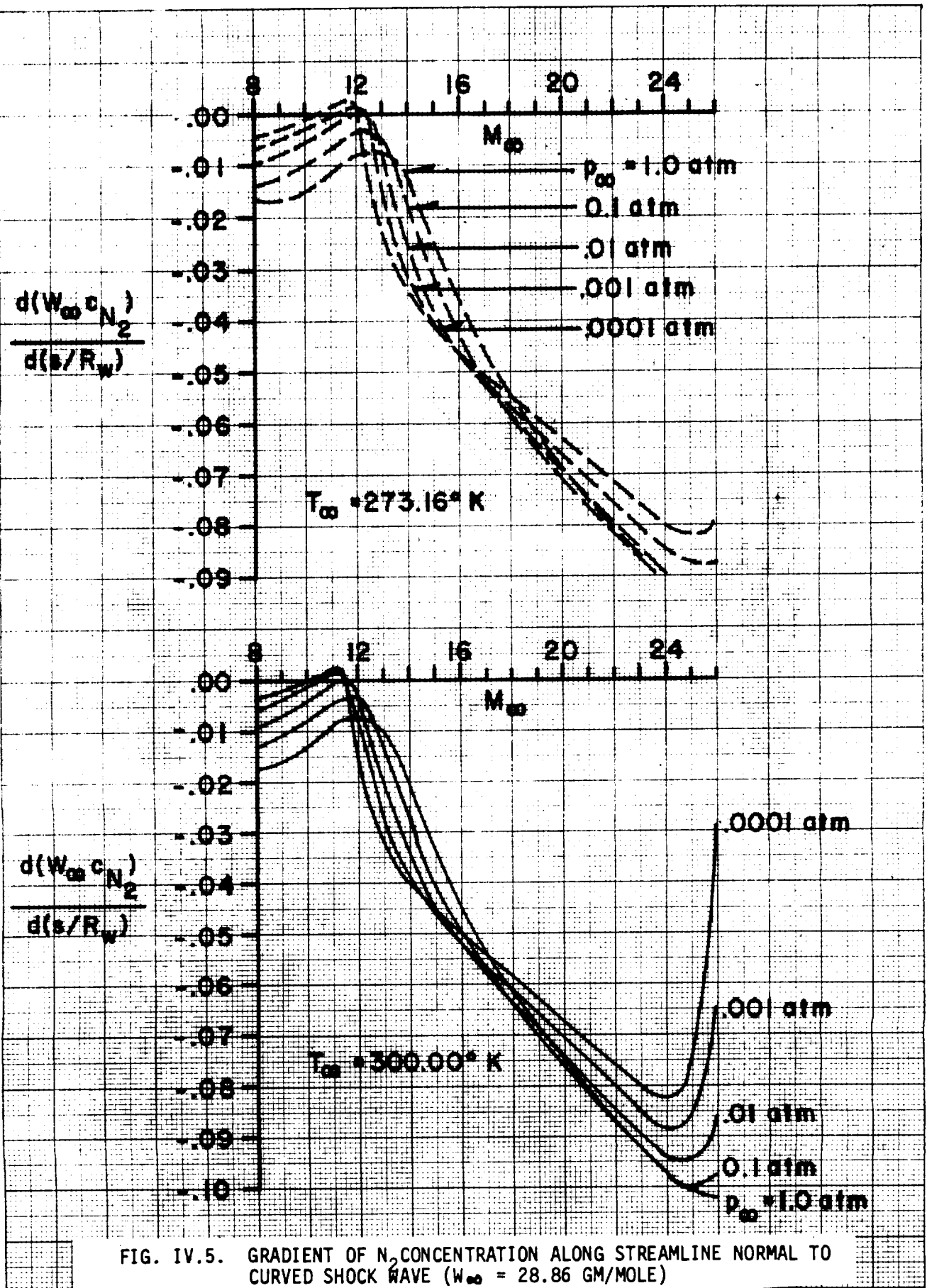


FIG. IV.5. GRADIENT OF  $N_2$  CONCENTRATION ALONG STREAMLINE NORMAL TO CURVED SHOCK WAVE ( $W_\infty = 28.86 \text{ GM/MOLE}$ )

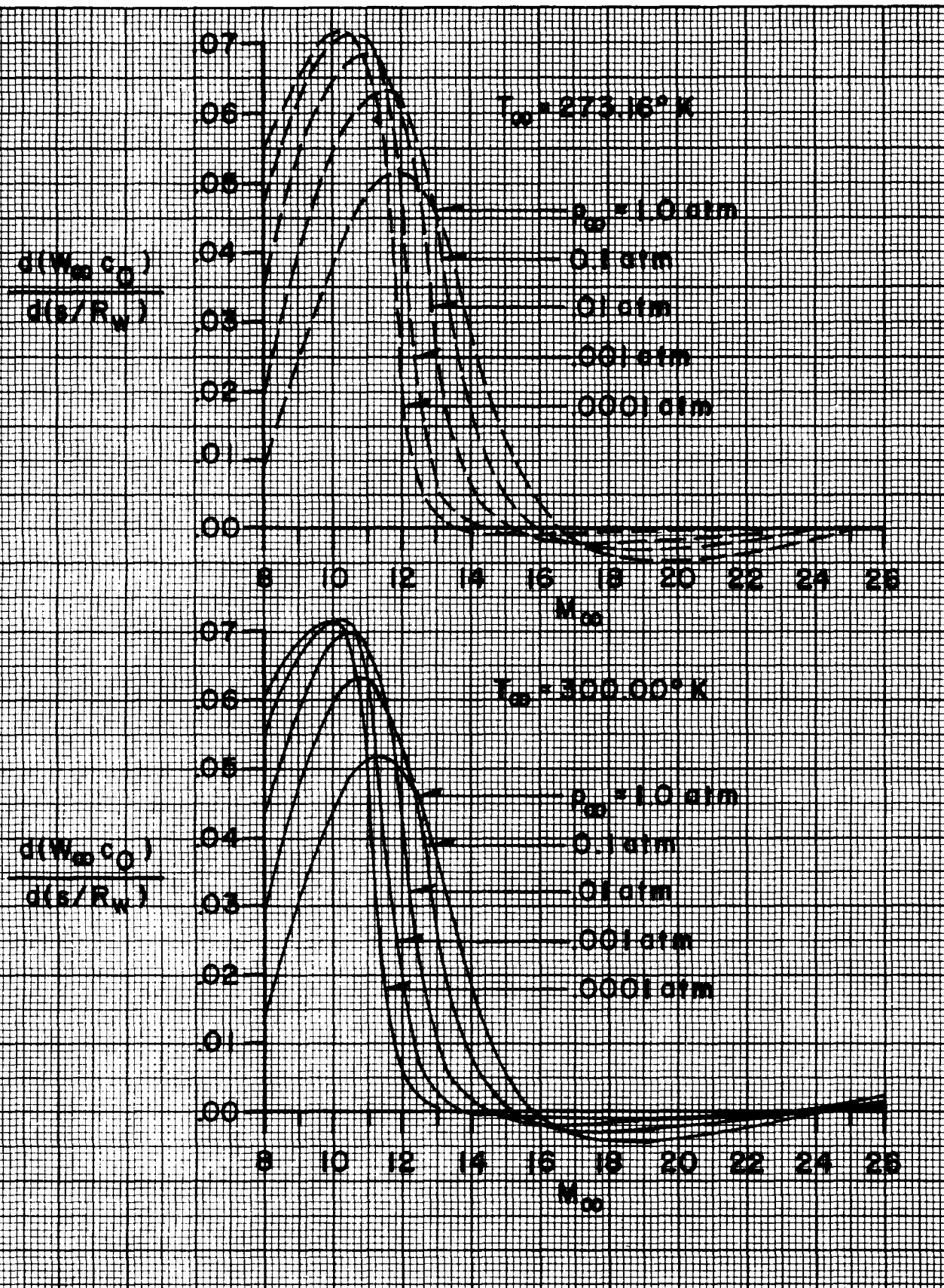


FIG. IV.6. GRADIENT OF O CONCENTRATION ALONG STREAMLINE NORMAL TO CURVED SHOCK WAVE ( $W_\infty = 28.86\text{ GM/MOLE}$ )

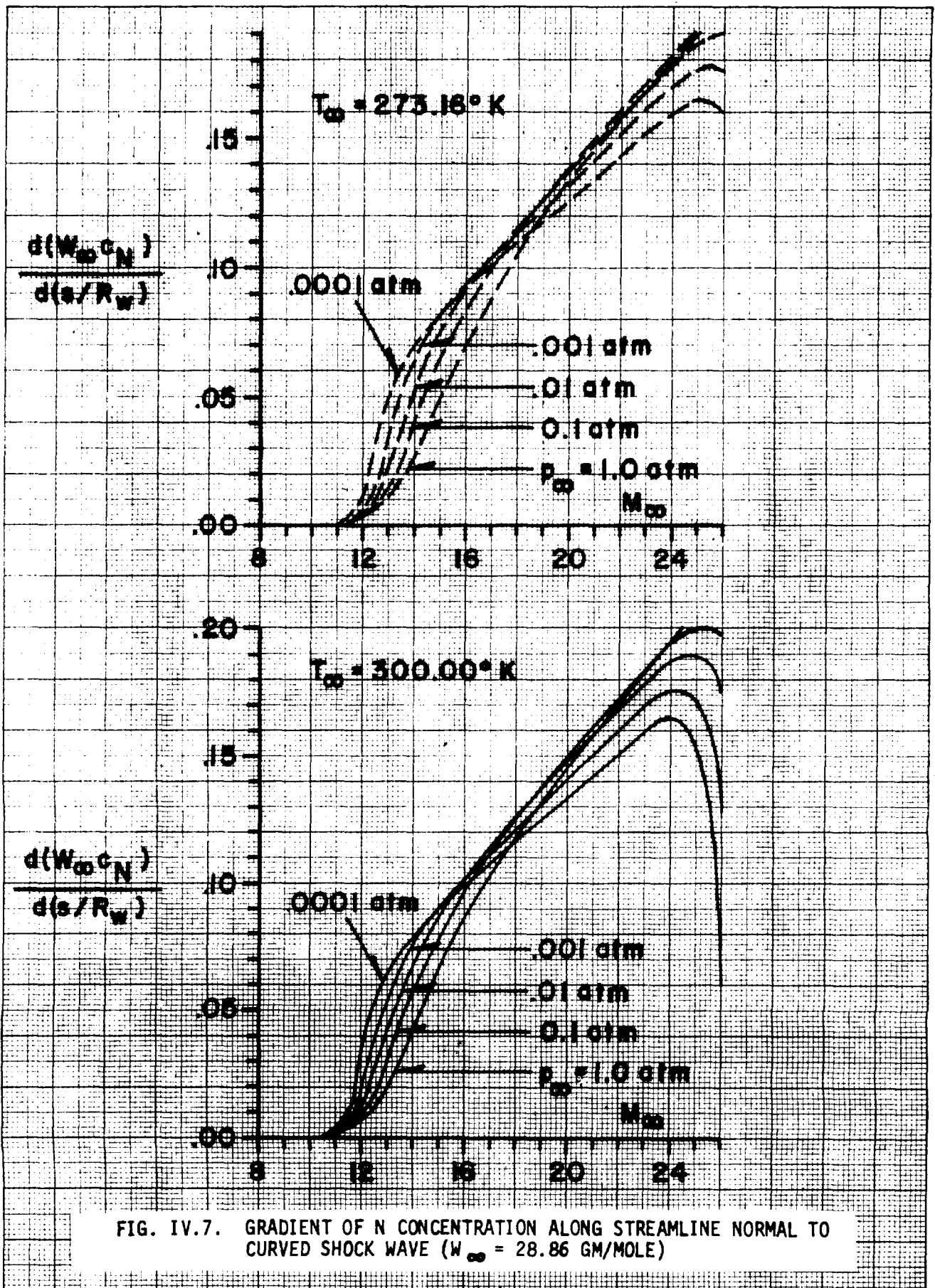


FIG. IV.7. GRADIENT OF N CONCENTRATION ALONG STREAMLINE NORMAL TO CURVED SHOCK WAVE ( $W_\infty = 28.86 \text{ GM/MOLE}$ )

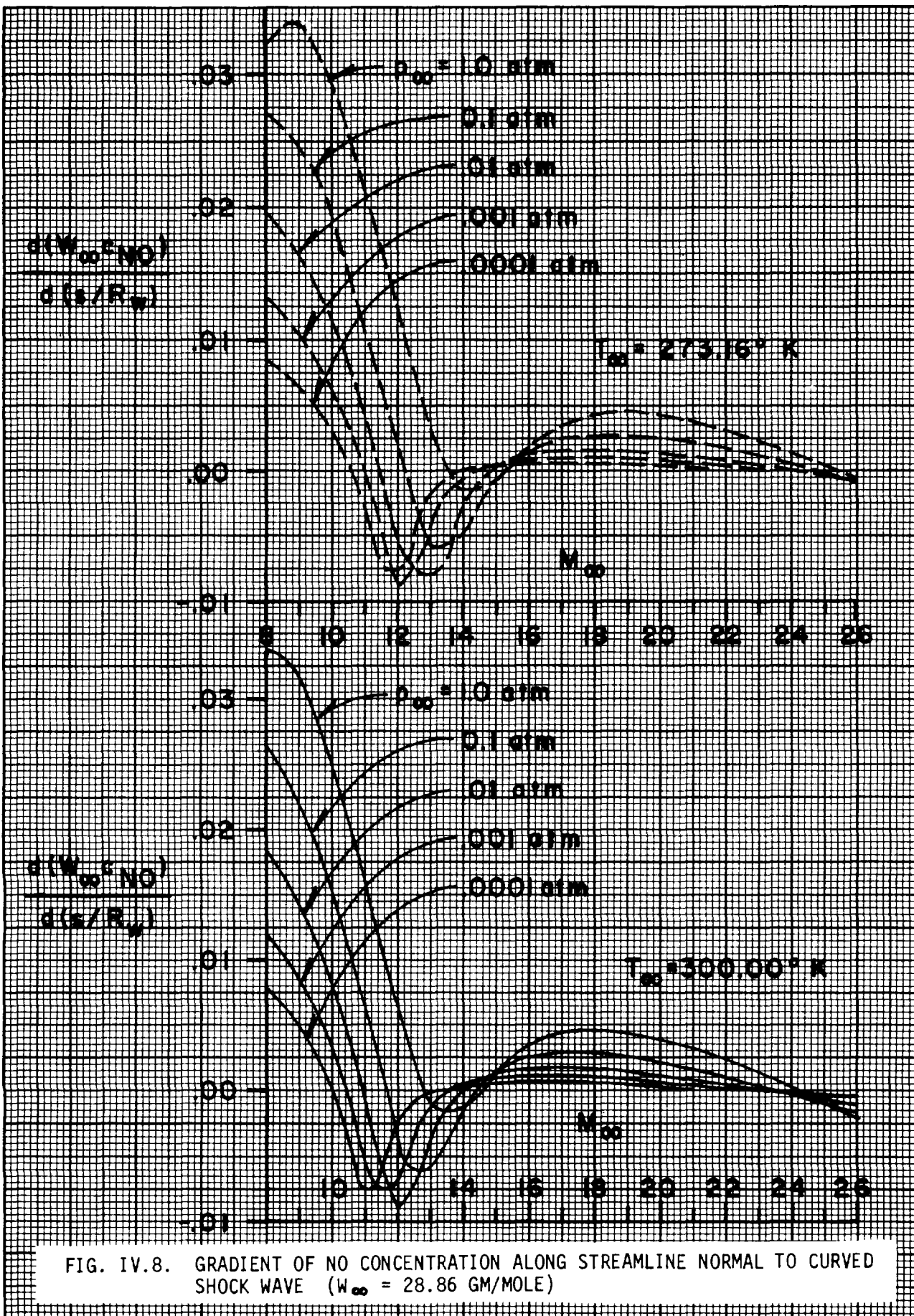


FIG. IV.8. GRADIENT OF NO CONCENTRATION ALONG STREAMLINE NORMAL TO CURVED SHOCK WAVE ( $W_\infty = 28.86 \text{ GM/MOLE}$ )

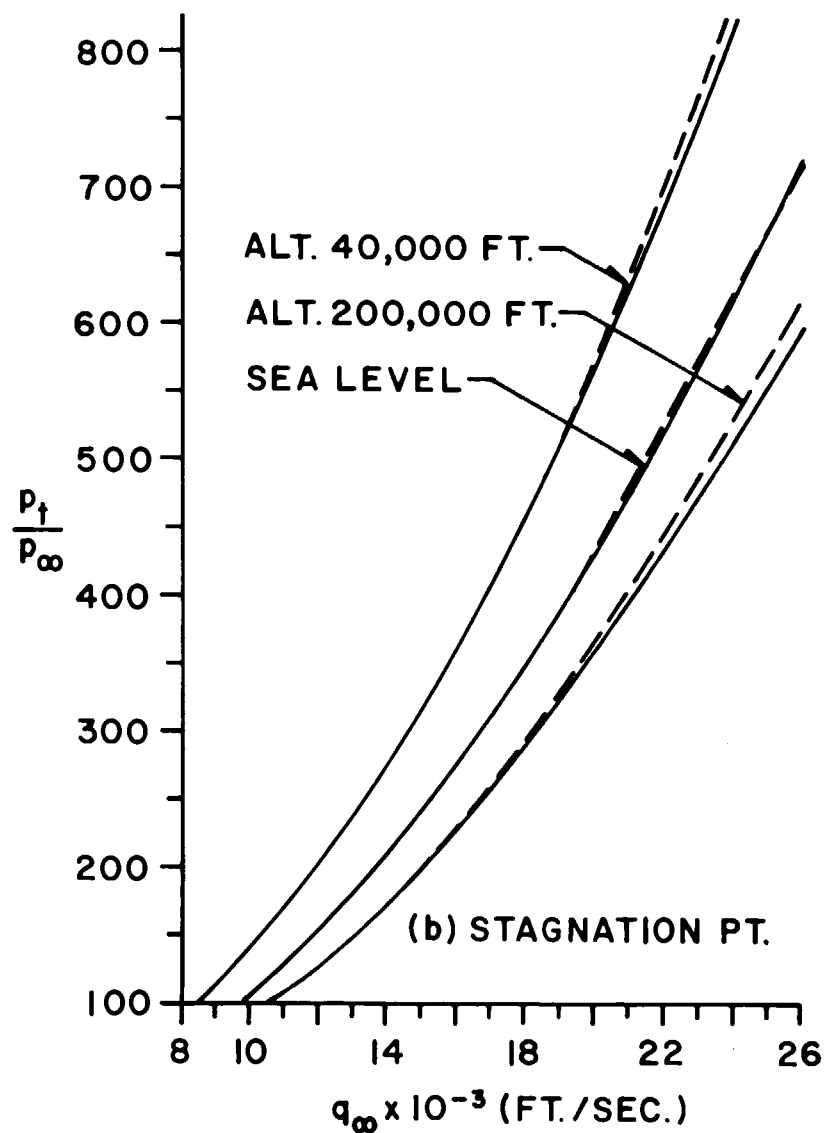
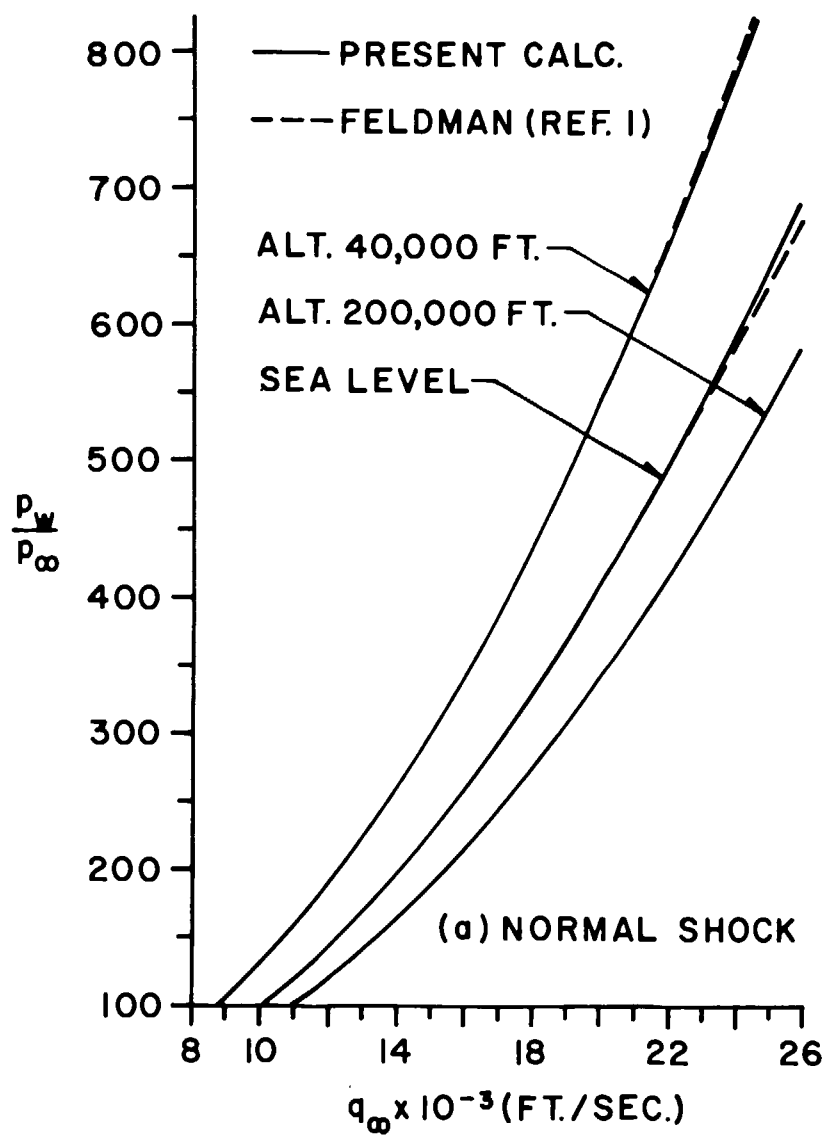


FIG. 7.1 COMPARISON OF PRESSURE CALCULATIONS WITH VALUES IN REF. 1.

## REFERENCES

1. Saul Feldman, "Hypersonic Gas Dynamic Charts for Equilibrium Air," Research Rept. 40, AVCO Research Lab (1957).
2. Henry E. Hudgins, Jr., "Supersonic Flow About Right Circular Cones at Zero Yaw in Air at Thermodynamic Equilibrium," T. M. 1493, Picatinny Arsenal, Dover, N. J. (1964).
3. Paul V. Marrone, "Normal Shock Waves in Air: Equilibrium Composition and Flow Parameters for Velocities from 26,000 to 50,000 Ft/Sec," CAL Rept. No. AG-1729-A-2, Cornell Aeronaut. Lab (1962).
4. Mary F. Romig, "Conical Flow Parameters for Air in Dissociation Equilibrium," Research Rept. 7, Convair Scientific Research Lab. (1960).
5. J. Spurk, N. Gerber, and R. Sedney, "Characteristic Calculation of Flow Fields with Chemical Reactions," Aberdeen Proving Ground, BRL R-1276 (March 1965). Also AIAA J., 4, 30-37 (1966).
6. Paul V. Marrone, "Inviscid, Nonequilibrium Flow Behind Bow and Normal Shock Waves, Part I. General Analysis and Numerical Examples," CAL Rept. No. QM-1626-A-12(I), Cornell Aeronaut. Lab. (1963).
7. K. L. Wray, "Chemical Kinetics of High Temperature Air," Progress in Astronautics and Rocketry, Vol. 7, Hypersonic Flow Research, F. R. Riddell (Editor), Academic Press, New York (1962).
8. Ames Research Staff, "Equations, Tables, and Charts for Compressible Flow," NACA R-1135 (1953).
9. A. Ralston and H. S. Wilf, Editors, Mathematical Methods for Digital Computers (p. 110), John Wiley and Sons, New York (1960).



APPENDIX  
COEFFICIENTS OF EQUATIONS (2.6) AND (2.9)

$$A_{11} = 2/\rho$$

$$A_{12} = A_{13} = 2/[c_0 + c_N + (2/W_\infty)]$$

$$A_{14} = 2/T$$

$$A_{21} = c_0 \left[ \frac{c_0}{K_1} + \frac{c_N}{2K_5} \right]$$

$$A_{22} = \frac{2\rho c_0}{K_1} + \frac{1}{2} + \frac{\rho c_N}{2K_5}$$

$$A_{23} = \frac{\rho c_0}{2K_5}$$

$$A_{24} = -\rho c_0 \left[ \frac{c_0}{K_1^2} \frac{dK_1}{dT} + \frac{c_N}{2K_5^2} \frac{dK_5}{dT} \right]$$

$$A_{31} = c_N \left[ \frac{c_N}{K_2} + \frac{c_0}{2K_5} \right]$$

$$A_{32} = \frac{\rho c_N}{2K_5}$$

$$A_{33} = \frac{2\rho c_N}{K_2} + \frac{1}{2} + \frac{\rho c_0}{2K_5}$$

$$A_{34} = -\rho c_N \left[ \frac{c_N}{K_2^2} \frac{dK_2}{dT} + \frac{c_0}{2K_5^2} \frac{dK_5}{dT} \right]$$

$$A_{41} = (c_0^2 h_{O_2}/K_1) + (c_N^2 h_{N_2}/K_2) + (c_N c_0 h_{NO}/K_5)$$

$$A_{42} = (2\rho c_0 h_{O_2}/K_1) + (\rho c_N h_{NO}/K_5) + h_{O_2}$$

$$A_{43} = (2\rho c_N h_{N_2}/K_2) + (\rho c_0 h_{NO}/K_5) + h_{N_2}$$

$$A_{44} = \sum_i c_i c_{p_i} - \rho \left[ \frac{c_0^2}{K_1^2} \frac{dK_1}{dT} h_{O_2} + \frac{c_N^2}{K_2^2} \frac{dK_2}{dT} h_{N_2} + \frac{c_0 c_N}{K_5^2} \frac{dK_5}{dT} h_{NO} \right]$$

TABLE I

## REACTION RATE COEFFICIENTS AND EQUILIBRIUM CONSTANTS

No. (j)	Reaction	Rate Coeff., Equilib. Const.	Catalyst( $\underline{M}$ )
I	$O_2 + \underline{M} \rightleftharpoons 2O + \underline{M}$	$(k_f)_1 = 1.2 \times 10^{18} T^{-3/2} \exp(-59,380/T)$ $K_1 = 1.2 \times 10^6 T^{-1/2} \exp(-59,380/T)$	$N_2, N, NO$
II	$N_2 + \underline{M} \rightleftharpoons 2N + \underline{M}$	$(k_f)_2 = 9.9 \times 10^{17} T^{-3/2} \exp(-113,260/T)$ $K_2 = 18.0 \times 10^3 \exp(-113,260/T)$	$O_2, O, NO$
III	$O_2 + O_2 \rightleftharpoons 2O + O_2$	$(k_f)_3 = 3.6 \times 10^{18} T^{-3/2} \exp(-59,380/T)$ $K_3 = \text{----same as for I}$	----
IV	$N_2 + N_2 \rightleftharpoons 2N + N_2$	$(k_f)_4 = 3.0 \times 10^{18} T^{-3/2} \exp(-113,260/T)$ $K_4 = \text{----same as for II}$	----
V	$NO + \underline{M} \rightleftharpoons N + O + \underline{M}$	$(k_f)_5 = 5.2 \times 10^{18} T^{-3/2} \exp(-75,490/T)$ $K_5 = 4.0 \times 10^3 \exp(-75,490/T)$	$O_2, O, N_2, N, NO$
VI	$O + N_2 \rightleftharpoons NO + N$	$(k_f)_6 = 5.0 \times 10^{10} \exp(-38,000/T)$ $K_6 = 4.5 \exp(-37,750/T)$	----
VII	$N + O_2 \rightleftharpoons NO + O$	$(k_f)_7 = 1.0 \times 10^9 T^{1/2} \exp(-3,120/T)$ $K_7 = 4.168687 \exp(+16,120/T)$	----
VIII	$O_2 + O \rightleftharpoons 3O$	$(k_f)_8 = 2.1 \times 10^{15} T^{-1/2} \exp(-59,380/T)$ $K_8 = \text{----same as for I}$	----
IX	$N_2 + N \rightleftharpoons 3N$	$(k_f)_9 = 1.5 \times 10^{19} T^{-3/2} \exp(-113,260/T)$ $K_9 = \text{----same as for II}$	----
X	$N_2 + O_2 \rightleftharpoons 2NO$	$(k_f)_{10} = 9.1 \times 10^{21} T^{-5/2} \exp(-65,000/T)$ $K_{10} = 19.0 \exp(-21,640/T)$	----

Dimensions:  $(k_f)_j$  --  $m^3/(kmol \text{ sec})$  ; $K_j$  for 3 body reactions --  $kmol/m^3$  .

TABLE II

## SPECIFIC ENTHALPIES AND SPECIFIC HEATS

$$h_{O_2} = 7RT/2 + (R\bar{\theta}_{O_2}) / [\exp(\bar{\theta}_{O_2}/T) - 1] + \bar{h}_{O_2}$$

$$h_{N_2} = 7RT/2 + (R\bar{\theta}_{N_2}) / [\exp(\bar{\theta}_{N_2}/T) - 1] + \bar{h}_{N_2}$$

$$h_O = 5RT/2 + \bar{h}_O$$

$$h_N = 5RT/2 + \bar{h}_N$$

$$h_{NO} = 7RT/2 + (R\bar{\theta}_{NO}) / [\exp(\bar{\theta}_{NO}/T) - 1] + \bar{h}_{NO}$$

$$(C_p)_{O_2} = 7R/2 + R(\bar{\theta}_{O_2}/T)^2 \exp(\bar{\theta}_{O_2}/T) / [\exp(\bar{\theta}_{O_2}/T) - 1]^2$$

$$(C_p)_{N_2} = 7R/2 + R(\bar{\theta}_{N_2}/T)^2 \exp(\bar{\theta}_{N_2}/T) / [\exp(\bar{\theta}_{N_2}/T) - 1]^2$$

$$(C_p)_O = 5R/2$$

$$(C_p)_N = 5R/2$$

$$(C_p)_{NO} = 7R/2 + R(\bar{\theta}_{NO}/T)^2 \exp(\bar{\theta}_{NO}/T) / [\exp(\bar{\theta}_{NO}/T) - 1]^2$$

No. (i)	Species	$\bar{\theta}_i (^{\circ}K)$	$\bar{h}_i$ (Dyn m/k mole)
1	O <sub>2</sub>	2256	0
2	N <sub>2</sub>	3374	0
3	O	--	2.467 65 x 10 <sup>8</sup>
4	N	--	4.710 63 x 10 <sup>8</sup>
5	NO	2719	0.898 655 x 10 <sup>8</sup>

1 Dyn. = 10<sup>5</sup> dynes

# DISTRIBUTION LIST

<u>No. of</u> <u>Copies</u>	<u>Organization</u>	<u>No. of</u> <u>Copies</u>	<u>Organization</u>
20	Commander Defense Documentation Center ATTN: TIPCR Cameron Station Alexandria, Virginia 22314	2	Commander U.S. Naval Ordnance Laboratory ATTN: Lib White Oak Silver Spring, Maryland 20910
1	Commanding General U.S. Army Materiel Command ATTN: Dir of Rsch & Labs Washington, D.C. 20315	1	Commander U.S. Naval Ordnance Test Station ATTN: Tech Lib China Lake, California 93557
1	Commanding General U.S. Army Materiel Command ATTN: AMCRD-RP-B Washington, D.C. 20315	1	Commander U.S. Naval Weapons Laboratory ATTN: Lib Dahlgren, Virginia 22448
2	Commanding General U.S. Army Missile Command ATTN: Tech Lib Redstone Arsenal, Alabama 35809	2	Superintendent U.S. Naval Postgraduate School ATTN: Tech Rpt Sec Monterey, California 93940
1	Commanding Officer U.S. Army Engineer Research & Development Laboratories ATTN: STINFO Div Fort Belvoir, Virginia 22060	1	AEDC (AER) Arnold AFS Tennessee 37389
1	Commanding General U.S. Army White Sands Missile Range White Sands Missile Range New Mexico 88002	1	RTD (RTTM) Bolling AFB Washington, D.C. 20332
1	Director U.S. Army Research Office 3045 Columbia Pike Arlington, Virginia 22204	1	APGC (PGBPS-12) Eglin AFB Florida 32542
3	Chief, Bureau of Naval Weapons ATTN: DLI-3 Department of the Navy Washington, D.C. 20360	1	AUL (3T-AUL-60-118) Maxwell AFB Alabama 36112
		1	Director Jet Propulsion Laboratory 4800 Oak Grove Drive Pasadena, California 91103
		1	Director Los Alamos Scientific Laboratory P.O. Box 1663 Los Alamos, New Mexico 87544

# DISTRIBUTION LIST

<u>No. of</u> <u>Copies</u>	<u>Organization</u>	<u>No. of</u> <u>Copies</u>	<u>Organization</u>
1	Sandia Corporation Livermore Branch ATTN: Tech Lib P.O. Box 969 Livermore, California 94551	1	Aerospace Corporation ATTN: Z. O. Bleviss P.O. Box 95085 Los Angeles, California 90045
1	Director Scientific and Technical Information Facility ATTN: NASA Rep (ATS) P.O. Box 5700 Bethesda, Maryland 20014	2	ARO, Inc. Arnold Engineering Development Center Gas Dynamics Facility ATTN: J. Lukasiewicz J. Whitfield P.O. Box 162 Tullahoma, Tennessee 37389
2	Director National Aeronautics and Space Administration Ames Research Center ATTN: Mr. H. J. Allen Mr. A. Seiff Moffett Field, California 94035	1	AVCO Corporation Research and Advanced Development Division ATTN: Lib 201 Lowell Street Wilmington, Massachusetts 01887
3	Director National Aeronautics and Space Administration Langley Research Center ATTN: Mr. Jerry South Mr. L. Roberts Mr. R. Trimpi Langley Station Hampton, Virginia 23365	1	AVCO-Everett Research Laboratory Advanced Development Division Research Laboratory ATTN: P. Rose 2385 Revere Beach Parkway Everett, Massachusetts 02149
1	Director National Aeronautics and Space Administration Lewis Research Center 21000 Brookpark Road Cleveland, Ohio 44135	1	The Beech Aircraft Corporation ATTN: Engr Lib Wichita, Kansas 67201
1	Applied Physics Laboratory The Johns Hopkins University 8621 Georgia Avenue Silver Spring, Maryland 20910	1	Boeing Airplane Company ATTN: T. Turner Seattle, Washington 98124
		2	Cornell Aeronautical Laboratory, Inc. ATTN: F. K. Moore J. G. Hall 4455 Genesee Street Buffalo, New York 14221

# DISTRIBUTION LIST

<u>No. of</u> <u>Copies</u>	<u>Organization</u>	<u>No. of</u> <u>Copies</u>	<u>Organization</u>
2	Douglas Aircraft Company, Inc. ATTN: Mr. R. J. Hakkinen Mr. C. Berend 3000 Ocean Park Boulevard Santa Monica, California 90405	1	Honeywell, Inc. Military Products Group 2345 Walnut Street St. Paul, Minnesota 55100
1	Electro-Optical Systems, Inc. 124 North Vinedo Avenue Pasadena, California 91100	2	IIT Research Institute ATTN: Mr. D. S. Hacker Mr. P. Chiarulli 10 West 35th Street Chicago, Illinois 60616
1	Firestone Tire and Rubber Company Defense Research Division 1200 Firestone Parkway Akron, Ohio 44317	1	Infrared Electro-Optical Co. 12 Skylane Drive Rochester, New York 14621
1	General Dynamics/CONVAIR ATTN: H. Yoshihara P.O. Box 1950 San Diego, California 92112	1	Ling-Temco-Vought, Inc. LTV Astronautics Division P.O. Box 5907 Dallas, Texas 75222
1	Ordnance Aerophysics Laboratory General Dynamics Corporation Daingerfield, Texas 75638	1	Arthur D. Little, Inc. 15 Acorn Park Cambridge, Massachusetts 02140
3	General Electric Company Missile and Space Division ATTN: Mr. W. Warren Mr. S. Scala Lib P.O. Box 80555 Philadelphia, Pennsylvania 19101	2	Lockheed Missiles and Space Company ATTN: Mr. D. Bershader Mr. R. Capiaux 3251 Hanover Street Palo Alto, California 94300
2	General Electric Company Aeronautics and Ordnance Systems Division ATTN: Mr. H. T. Nagamatsu Mr. D. R. White 1 River Road Schenectady, New York 13205	2	The Martin Company ATTN: Mr. J. Sternberg Mr. S. Moslin Baltimore, Maryland 21203
1	General Motors Corporation Defense Research Laboratories ATTN: Dr. A. C. Charters Santa Barbara, California 93108	1	North American Aviation, Inc. Aerophysics Laboratory 12214 Lakewood Boulevard Downey, California 90241
		1	Northrop Corporation Noriar Division 1001 East Broadway Hawthorne, California 90250

# DISTRIBUTION LIST

<u>No. of Copies</u>	<u>Organization</u>	<u>No. of Copies</u>	<u>Organization</u>
2	Philco Corporation Aeronautic Division Ford Road Newport Beach, California 92600	1	Yale University Mason Laboratory ATTN: W. A. Klikoff, Jr. 400 Temple Street New Haven, Connecticut 06510
1	Ramo-Wooldridge Corporation 5730 Arbor Vitae Street Los Angeles, California 90045	1	Professor M. L. Bactor Department of Engineer Graphics Iowa State University of Science and Technology Ames, Iowa 50010
1	Space Technology Laboratories Inc. ATTN: Mr. J. R. Kliegel P.O. Box 95001 Los Angeles, California 90045	1	Professor S. Bogdonoff Forrestal Research Center Princeton University Princeton, New Jersey 08540
1	Massachusetts Institute of Technology ATTN: Mr. J. Baron Cambridge, Massachusetts 02139	1	Professor G. F. Carrier Division of Engineering and Applied Physics Harvard University Cambridge, Massachusetts 01938
2	Polytechnic Institute of Brooklyn ATTN: Mr. Martin Bloom 527 Atlantic Avenue Freeport, New York 11520	1	Professor F. H. Clauser, Jr. Department of Aeronautics The Johns Hopkins University Baltimore, Maryland 21218
2	Rensselaer Polytechnic Institute ATTN: Mr. T. Y. Li Mr. G. H. Handelman Troy, New York 12181	1	Professor H. W. Emmons Harvard University Cambridge, Massachusetts 01238
1	Stanford University ATTN: Mr. M. D. VanDyke Stanford, California 94305	1	Professor W. D. Haynes Department of Aeronautical Engineering Forrestal Research Center Princeton University Princeton, New Jersey 08540
1	University of Illinois Aeronautical Institute Urbana, Illinois 61803		
1	University of Southern California ATTN: Dir, Engr Ctr Los Angeles, California 90007	1	Professor L. Lees Guggenheim Aeronautical Lab California Institute of Technology Pasadena, California 91004



## DISTRIBUTION LIST

<u>No. of</u> <u>Copies</u>	<u>Organization</u>
1	Professor H. W. Leipmann Aeronautics Department California Institute of Technology 1201 East California Blvd Pasadena, California 91102
1	Professor S. I. Pai Institute of Fluid Dynamics and Applied Mathematics University of Maryland College Park, Maryland 20740
1	Professor R. F. Probst Massachusetts Institute of Technology Cambridge, Massachusetts 02139
1	Dr. J. Clark Division of Applied Mathematics Brown University Providence, Rhode Island 02912
1	Dr. A. E. Puckett Systems Development Laboratory Hughes Aircraft Company Florence Avenue at Teal Street Culver City, California 90232
1	Dr. W. R. Sears Graduate School of Aeronautical Engineering Cornell University Ithaca, New York 14850

### Aberdeen Proving Ground

Ch, Tech Lib

Air Force Ln Ofc  
Marine Corps Ln Ofc  
Navy Ln Ofc  
CDC Ln Ofc

## DOCUMENT CONTROL DATA - R&amp;D

(Security classification of title, body of abstract and indexing annotation must be entered when the overall report is classified)

1. ORIGINATING ACTIVITY (Corporate author) U.S. Army Ballistic Research Laboratories Aberdeen Proving Ground, Maryland		2a. REPORT SECURITY CLASSIFICATION Unclassified	
		2b. GROUP	
3. REPORT TITLE CALCULATIONS FOR AIR FLOWS IN DISSOCIATION EQUILIBRIUM			
4. DESCRIPTIVE NOTES (Type of report and inclusive dates)			
5. AUTHOR(S) (Last name, first name, initial) Gerber, Nathan and Bartos, Joan M.			
6. REPORT DATE November 1965		7a. TOTAL NO. OF PAGES 65	7b. NO. OF REFS 9
8a. CONTRACT OR GRANT NO.		9a. ORIGINATOR'S REPORT NUMBER(S) Report No. 1306	
b. PROJECT NO. 1A222901A201			
c.		9b. OTHER REPORT NO(S) (Any other numbers that may be assigned this report)	
d.			
10. AVAILABILITY/LIMITATION NOTICES Distribution of this document is unlimited.			
11. SUPPLEMENTARY NOTES		12. SPONSORING MILITARY ACTIVITY U.S. Army Materiel Command Washington, D.C.	
13. ABSTRACT <p>Results of calculations carried out for a model of air in dissociation equilibrium are presented in graphical form. The quantities computed are i) flow variables (including species concentrations) behind normal and oblique shock waves, ii) flow variables in axisymmetric conical flow fields, iii) stagnation point values of flow variables on the 'stagnation' streamline behind two-dimensional and axisymmetric detached shock waves, and iv) flow variable gradients at the shock wave on stagnation streamlines. Computations are given for free stream temperatures of 273.16°K and 300°K, free stream pressures of 1.0, .1, .01, .001, and .0001 atmospheres, and a range of initial Mach numbers and cone angles to provide flow field temperatures in the range 3000°K - 10,000°K. Brief derivations of the equations employed are given.</p> <p>The present calculations are oriented toward application in experiments in hyper-sonic flow with ground facilities such as shock tubes and ballistic ranges. In addition, they furnish important supplementary information to theoretical studies of nonequilibrium flows.</p>			

14. KEY WORDS	LINK A		LINK B		LINK C	
	ROLE	WT	ROLE	WT	ROLE	WT
Hypersonic flow Dissociation (chemical reactions) Equilibrium flow Computation (high speed) Aerodynamic theory Air (multicomponent gas) Fluid dynamics Shock waves Supersonic flow Stagnation point						

**INSTRUCTIONS**

**1. ORIGINATING ACTIVITY:** Enter the name and address of the contractor, subcontractor, grantee, Department of Defense activity or other organization (*corporate author*) issuing the report.

**2a. REPORT SECURITY CLASSIFICATION:** Enter the overall security classification of the report. Indicate whether "Restricted Data" is included. Marking is to be in accordance with appropriate security regulations.

**2b. GROUP:** Automatic downgrading is specified in DoD Directive 5200.10 and Armed Forces Industrial Manual. Enter the group number. Also, when applicable, show that optional markings have been used for Group 3 and Group 4 as authorized.

**3. REPORT TITLE:** Enter the complete report title in all capital letters. Titles in all cases should be unclassified. If a meaningful title cannot be selected without classification, show title classification in all capitals in parenthesis immediately following the title.

**4. DESCRIPTIVE NOTES:** If appropriate, enter the type of report, e.g., interim, progress, summary, annual, or final. Give the inclusive dates when a specific reporting period is covered.

**5. AUTHOR(S):** Enter the name(s) of author(s) as shown on or in the report. Enter last name, first name, middle initial. If military, show rank and branch of service. The name of the principal author is an absolute minimum requirement.

**6. REPORT DATE:** Enter the date of the report as day, month, year; or month, year. If more than one date appears on the report, use date of publication.

**7a. TOTAL NUMBER OF PAGES:** The total page count should follow normal pagination procedures, i.e., enter the number of pages containing information.

**7b. NUMBER OF REFERENCES:** Enter the total number of references cited in the report.

**8a. CONTRACT OR GRANT NUMBER:** If appropriate, enter the applicable number of the contract or grant under which the report was written.

**8b, 8c, & 8d. PROJECT NUMBER:** Enter the appropriate military department identification, such as project number, subproject number, system numbers, task number, etc.

**9a. ORIGINATOR'S REPORT NUMBER(S):** Enter the official report number by which the document will be identified and controlled by the originating activity. This number must be unique to this report.

**9b. OTHER REPORT NUMBER(S):** If the report has been assigned any other report numbers (*either by the originator or by the sponsor*), also enter this number(s).

**10. AVAILABILITY/LIMITATION NOTICES:** Enter any limitations on further dissemination of the report, other than those imposed by security classification, using standard statements such as:

- (1) "Qualified requesters may obtain copies of this report from DDC."
- (2) "Foreign announcement and dissemination of this report by DDC is not authorized."
- (3) "U. S. Government agencies may obtain copies of this report directly from DDC. Other qualified DDC users shall request through \_\_\_\_\_."
- (4) "U. S. military agencies may obtain copies of this report directly from DDC. Other qualified users shall request through \_\_\_\_\_."
- (5) "All distribution of this report is controlled. Qualified DDC users shall request through \_\_\_\_\_."

If the report has been furnished to the Office of Technical Services, Department of Commerce, for sale to the public, indicate this fact and enter the price, if known.

**11. SUPPLEMENTARY NOTES:** Use for additional explanatory notes.

**12. SPONSORING MILITARY ACTIVITY:** Enter the name of the departmental project office or laboratory sponsoring (*paying for*) the research and development. Include address.

**13. ABSTRACT:** Enter an abstract giving a brief and factual summary of the document indicative of the report, even though it may also appear elsewhere in the body of the technical report. If additional space is required, a continuation sheet shall be attached.

It is highly desirable that the abstract of classified reports be unclassified. Each paragraph of the abstract shall end with an indication of the military security classification of the information in the paragraph, represented as (TS), (S), (C), or (U).

There is no limitation on the length of the abstract. However, the suggested length is from 150 to 225 words.

**14. KEY WORDS:** Key words are technically meaningful terms or short phrases that characterize a report and may be used as index entries for cataloging the report. Key words must be selected so that no security classification is required. Identifiers, such as equipment model designation, trade name, military project code name, geographic location, may be used as key words but will be followed by an indication of technical context. The assignment of links, rules, and weights is optional.